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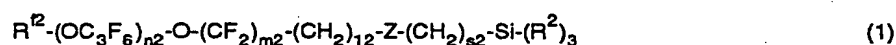
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(54) Anti-smudge agent, method of forming an anti-smudge layer, anti-reflection optical member and display device

(57) An anti-smudge agent characterized by containing an organosilicon compound represented by the following general formula (1):



wherein  $R^2$  is a linear or branched perfluoroalkyl group having 1 to 16 carbon atoms;  $R^2$  is a hydrolytic group; Z is  $-OOCNH-$  or  $-O-$ ;  $n2$  is an integer of 1 to 50;  $m2$  is an integer of 0 to 3; 12 is an integer of 0 to 3; and  $s2$  is an integer of 1 to 6, with a proviso that these  $m2$  and 12 meet the condition of  $6 \geq m2 + 12 > 0$ .

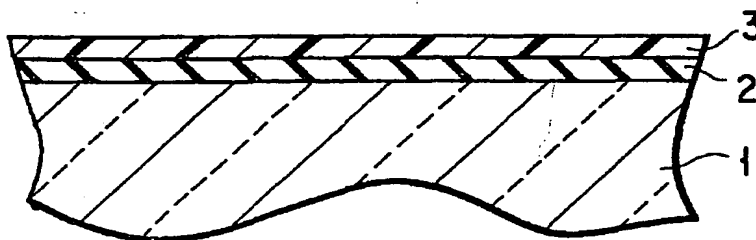


FIG. 1

## Description

[0001] This invention relates to an anti-smudge agent for forming an anti-smudge layer on the surface of various kinds of substrate to be treated (hereinafter referred to simply as a treating substrate) and requiring an anti-smudge property, and to a method of forming an anti-smudge layer by making use of this anti-smudge agent. Additionally, this invention relates to an anti-reflection optical member which is adapted to be applied to the surface of display screen of an optical member (such as an anti-reflection film, an optical filter, an optical lens, spectacle lens, beam splitter, prism, mirror, etc.) and a device (such as a liquid crystal display, a CRT display, a projection television, a plasma display, an EL display, etc.), to an optical functional member, and to a display device wherein this optical functional member is adhered to the surface of display screen of the display device.

[0002] Optical members such as an anti-reflection film, an optical filter, an optical lens, spectacle lens, beam splitter, prism, mirror, etc. are liable to be soiled with smudges originating from fingerprints, skin fat, sweat, cosmetics, etc. as these members are employed by an operator. Once these members are soiled with these smudges, it is very difficult to remove them. Particularly in the case of an optical member provided with an antireflection coating, the smudge adhered to the antireflection coating stands out from the coating, thus raising a problem.

[0003] Meanwhile, most of display devices are generally employed under the condition that external light enters freely into the display screen from the inside as well as outside of the room.

[0004] The incident light like this external light is mirror-reflected by the surface of the display screen, thus causing a mixing of images between the reflected image and the displayed image, resulting in the deterioration of the display quality and hence the blurring of the display image.

[0005] In particular, since the frequency of using a computer has been recently increased with an progress of office automation, and hence the time during which an operator stares at the display screen has been also increased recently, the deterioration of display quality due to the reflection image is now considered as being one of the causes for bringing about a health hazard such as the fatigue of eyes.

[0006] Furthermore, with an increasing popularization of out-door life in recent years, the opportunity of using various kinds of display device outside the door is increasing now. As a result, it is now demanded to improve the quality of display so as to enable a display image to be recognized more clearly.

[0007] With a view to meet these demands, there has been conventionally proposed an idea of attaching a reflection preventive optical member to the surface of the display screen, this reflection preventive optical member being capable of preventing the reflection of light over a wide range of visible light. One example of such a reflection preventive optical member is a laminate film comprising a transparent substrate on the surface which a high refractive index layer and a low refractive index layer both consisting of a metal oxide are laminated. There is also known a reflection preventive optical member which is formed of a laminate film comprising a transparent substrate on the surface which a low refractive index layer consisting of an inorganic or organic fluorine compound is laminated.

[0008] On the other hand, it is also known that the same effect as that of a reflection preventive optical member can be obtained by the employment of an irregular reflective film comprising a transparent plastic film substrate on the surface of which a coating layer containing transparent fine particles is deposited to form a rugged surface so as to cause the external light to be irregularly reflected by the rugged surface.

[0009] In the meantime, the aforementioned reflection preventive optical members tend to be soiled with smudges originating from fingerprints, skin fat, sweat, cosmetics, etc. as the display device provided with this reflection preventive optical member is employed by an operator. Specifically, since the surface energy of the reflection preventive film is generally as large as 60 J/m<sup>2</sup> or so, the reflection preventive film is more likely to be soiled with these smudges. Further, since the surface of the reflection preventive film is constituted by a finely roughened surface, once these smudges are adhered onto the reflection preventive film, it is very difficult to remove them. Further, there is also a problem that when these smudges are adhered to a portion of the reflection preventive film, the soiled portion becomes more reflective, thereby rendering the soiled portion to stand out from the remaining portion.

[0010] With a view to solve this problem of the smudge on the surface of reflection preventive optical member, various techniques for forming an anti-smudge layer which is capable of preventing smudges from adhering thereon and capable of easily removing the adhered smudges have been proposed up to date.

[0011] Japanese Patent Unexamined Publication S64-86101 teaches a reflection preventive article having anti-smudge and abrasion resistive properties wherein the surface of the substrate is covered with a reflection preventive film consisting mainly of silicon dioxide, and the resultant surface is further treated with a compound containing an organosilicic substituent group. Japanese Patent Unexamined Publication H4-338901 teaches a CRT filter having anti-smudge and abrasion resistive properties, whose substrate surface is covered with a terminal silanol organopolysiloxane. Japanese Patent Publication H6-29332 teaches an anti-smudge/low reflectance plastic covered on its surface with a reflection preventive film containing mono- and di-silane compounds containing polyfluoroalkyl group, and a silanized halogen, alkyl or alkoxy compound. Further, Japanese Patent Unexamined Publication H7-16940 teaches an optical member comprising an optical thin film consisting mainly of silicon dioxide, on which a copolymer consisting

of perfluoroalkyl(meta)acrylate and a monomer having alkoxy silane group is formed.

**[0012]** However, the anti-smudge layer manufactured according to these conventional methods is insufficient in anti-smudge property, incapable of allowing a smudge such as fingerprints, skin fat, sweat, cosmetics, etc. to be easily removed therefrom, and poor in durability losing sharply its anti-smudge property with a repetition of use. Therefore, it is now desired to develop an anti-smudge layer which is excellent in anti-smudge property and in the durability of anti-smudge property.

**[0013]** This invention has been made to overcome the aforementioned conventional problems, and therefore, an object of this invention is to provide an anti-smudge agent, which makes it possible to obtain an anti-smudge layer which is excellent in durability of the property, capable of preventing smudges formed of fingerprints, skin fat, sweat, cosmetics, etc. from being adhered onto the surface of an optical member such as a reflection preventive film, and capable of allowing these smudges to be easily removed even if these smudges are adhered onto the optical member.

**[0014]** Another object of this invention is to provide a method of forming an anti-smudge layer which is capable of easily forming an anti-smudge layer which is excellent in anti-smudge property and in the durability of the anti-smudge property.

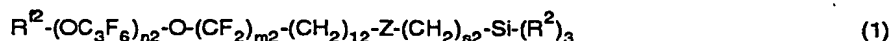
**[0015]** Still another object of this invention is to provide an optical member which is provided with the aforementioned anti-smudge layer.

**[0016]** Still another object of this invention is to provide a reflection preventive optical member which is provided with the aforementioned anti-smudge layer.

**[0017]** A further object of this invention is to provide an optical functional member which is provided with the aforementioned reflection preventive optical member.

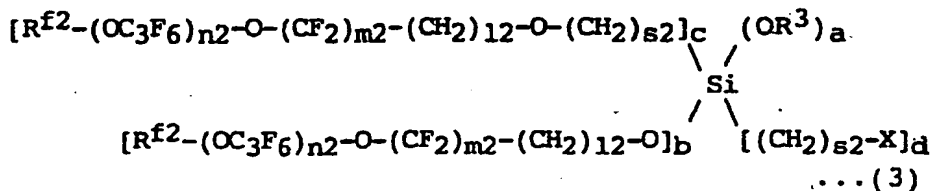
**[0018]** A still further object of this invention is to provide a display device whose display screen surface is constituted by the aforementioned optical functional member.

**[0019]** According to this invention, there is provided an anti-smudge agent containing an organosilicon compound represented by the following general formula (1):



wherein  $R^2$  is a linear or branched perfluoroalkyl group having 1 to 16 carbon atoms;  $F^2$  is a hydrolytic group; Z is -OOCNH- or -O-;  $n_2$  is an integer of 1 to 50;  $m_2$  is an integer of 0 to 3; 12 is an integer of 0 to 3; and  $s_2$  is an integer of 1 to 6, with a proviso that these  $m_2$  and 12 meet the condition of  $6 \geq m_2 + 12 > 0$ .

**[0020]** According to this invention, there is provided an anti-smudge agent containing an organosilicon compound represented by the following general formula (3):



wherein  $R^2$  is a linear or branched perfluoroalkyl group having 1 to 16 carbon atoms;  $R^3$  is an alkyl group having 1 to 10 carbon atoms; X is a halogen atom; n2 is an integer of 1 to 50; m2 is an integer of 0 to 3; 12 is an integer of 0 to 3; and s2 is an integer of 1 to 6; a is an integer of 0 to 3; b is an integer of 0 to 3; c is 0 or 1; d is 0 or 1; with a proviso that these m2, 12, a, b, c and d meet the conditions of  $6 \geq m2 + 12 > 0$ ,  $a + b = 3$ , and  $c + d = 1$ .

**[0021]** According to this invention, there is further provided a method of forming an anti-smudge layer containing the aforementioned anti-smudge agent.

**[0022]** According to this invention, there is further provided an optical member provided with an anti-smudge layer containing the aforementioned anti-smudge agent.

**[0023]** According to this invention, there is further provided a reflection preventive optical member provided with an anti-smudge layer containing the aforementioned anti-smudge agent.

**[0024]** According to this invention, there is further provided an optical functional member constituted by the aforementioned reflection preventive optical member.

**[0025]** According to this invention, there is further provided a display device provided with the aforementioned optical functional member.

[0026] This summary of the invention does not necessarily describe all necessary features so that the invention may also be a sub-combination of these described features.

[0027] The invention can be more fully understood from the following detailed description when taken in conjunction with the accompanying drawings, in which:

FIG. 1 is a cross-sectional view illustrating the structure of one example of reflection preventive optical member according to this invention;

FIG. 2 is a cross-sectional view illustrating the structure of one example of optical functional member according to this invention;

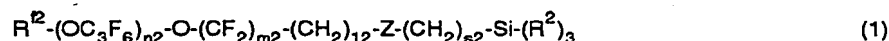
FIGS. 3A and 3B are schematical views each illustrating the structure of one example of display device according to this invention;

FIG. 4 is an FT-IR chart of an organosilicon compound constituting an anti-smudge agent according to this invention; and

FIG. 5 is a cross-sectional view illustrating the structure of another example of reflection preventive optical member according to this invention.

[0028] The anti-smudge agent according to this invention is characterized in that it contains a specific kind of organosilicon compound. As a result, when an anti-smudge layer is formed on the surface of a treating substrate by making use of this anti-smudge agent, it is possible, if the treating substrate is any kind of optical members such as an anti-reflection film, an optical filter, an optical lens, spectacle lens, beam splitter, prism, mirror, etc., to prevent smudges formed any of fingerprints, skin fat, sweat, cosmetics, etc. from being adhered onto the surface of these optical members, and furthermore, to easily remove these smudges adhered onto these optical members without deteriorating the optical properties of these optical members. Moreover, the anti-smudge layer formed by making use of this anti-smudge agent is excellent in durability of the property.

[0029] Namely, the anti-smudge agent according to this invention contains an organosilicon compound represented by the following general formula (1):



wherein  $R^f$  is a linear or branched perfluoroalkyl group having 1 to 16 carbon atoms;  $R^2$  is a hydrolytic group; Z is -OOCNH- or -O-;  $n2$  is an integer of 1 to 50;  $m2$  is an integer of 0 to 3; 12 is an integer of 0 to 3; and  $s2$  is an integer of 1 to 6, with a proviso that these  $m2$  and 12 meet the condition of  $6 \geq m2 + 12 > 0$ .

[0030] The number of carbon atom of the linear or branched perfluoroalkyl group to be introduced into  $R^f$  is preferably in the range of from 1 to 16, more preferably in the range of from 1 to 3. Therefore, most preferable examples of  $R^f$  are  $-CF_3$ ,  $-C_2F_5$ ,  $-C_3F_7$ , etc.

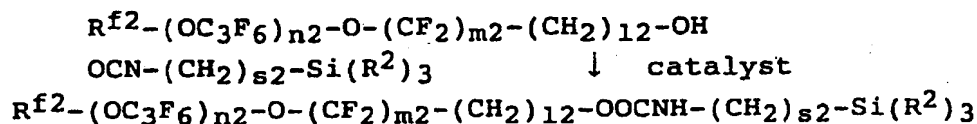
[0031] Specific examples of the hydrolytic group to be introduced into  $R^2$  are -Cl, -Br, -I, -OR<sup>11</sup>, -OOCR<sup>11</sup>, -OC(R<sup>11</sup>)C=C(R<sup>12</sup>)<sub>2</sub>, -ON=C(R<sup>11</sup>)<sub>2</sub>, -ON=CR<sup>13</sup>, -N(R<sup>12</sup>)<sub>2</sub>, -R<sup>12</sup>NOCR<sup>11</sup>, etc., wherein R<sup>11</sup> is an aliphatic hydrocarbon group having 1 to 10 carbon atoms such as alkyl group or an aromatic hydrocarbon group having 6 to 20 carbon atoms such as phenyl group; R<sup>12</sup> is hydrogen atom or an aliphatic hydrocarbon group having 1 to 5 carbon atoms such as alkyl group; and R<sup>13</sup> is a bivalent aliphatic hydrocarbon group having 3 to 6 carbon atoms such as alkylidene.

[0032] Most preferable examples of the hydrolytic group are -OCH<sub>3</sub>, -OC<sub>2</sub>H<sub>5</sub>, -OOCCH<sub>3</sub>, and -NH<sub>2</sub>.

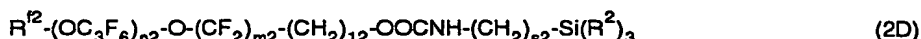
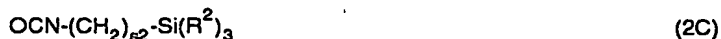
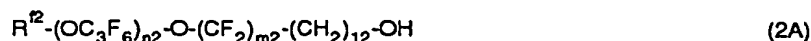
[0033] These hydrolytic groups may be included singly or in combination of two or more kinds in the organosilicon compound constituting the anti-smudge agent of this invention.

[0034] The  $n2$  in the aforementioned general formula (1) should preferably be in the range of 1 to 30, more preferably in the range of 5 to 20. The  $m2$  in the aforementioned general formula (1) should preferably be in the range of 1 to 2. The 12 in the aforementioned general formula (1) should preferably be in the range of 1 to 3.

[0035] The compounds where Z is -OOCNH- in the general formula (1) may be prepared according to the following reaction formula:



[0036] More specifically, an organosilicon compound represented by the following general formula (2D) can be obtained by allowing a fluorine resin having hydroxyl group and represented by the following general formula (2A) to react with catalyst and a isocyanate-based silane coupling agent represented by the following general formula (2C) in a solvent.



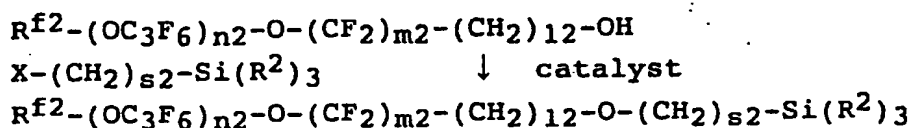
wherein  $R^{f2}$  is a linear or branched perfluoroalkyl group having 1 to 16 carbon atoms;  $R^2$  is a hydrolytic group;  $n2$  is an integer of 1 to 50;  $m2$  is an integer of 0 to 3; 12 is an integer of 0 to 3; and  $s2$  is an integer of 0 to 6.

[0037] This reaction can be advantageously proceeded by using 0.001 to 0.01 mol of  $C_{32}H_{64}O_4Sn$  as a catalyst and 1 to an excess of a isocyanate-based silane coupling agent per 1 mol of a fluorine resin containing hydroxyl group and by performing the reaction at a temperature ranging from room temperature to the reflux temperature of the solvent for 2 to 72 hours. As for the solvent useful in this case, it is possible to employ an ordinary perfluoroaliphatic hydrocarbon having 5 to 12 carbon atoms, such as perfluorohexane, perfluoromethylcyclohexane, perfluoro-1,3-dimethylcyclohexane, etc.; an aromatic hydrocarbon polyfluoride such as bis(trifluoromethyl) benzene; or an aliphatic hydrocarbon polyfluoride.

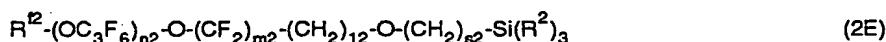
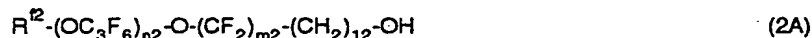
[0038] The anti-smudge agent according to this invention may include, if desired, various kinds of additives in addition to the aforementioned organometallic compounds represented by the general formula (1). For example, the anti-smudge agent may contain a catalyst for promoting the hydrolysis or polycondensation of the organosilicon compounds represented by the general formula (1) or for promoting the formation of condensed urethane linkage.

[0039] As to the catalyst that can be employed in the anti-smudge agent of this invention, it is possible to employ hydrochloric acid, nitric acid, sulfuric acid, acetic acid, hydrofluoric acid, formic acid, phosphoric acid, oxalic acid, ammonia, aluminum acetylacetonate, dibutyltin dilaurate, octylate tin compound, sodium, sodium hydride, methane sulfonate, trifluoromethane sulfonate, paratoluene sulfonate, trifluoroacetate, etc. These compounds can be employed singly or in combination of two or more kinds.

[0040] The compounds where Z is -O- in the general formula (1) may be prepared according to the following reaction formula:



[0041] More specifically, an organosilicon compound represented by the following general formula (2E) can be obtained by allowing a fluorine resin having hydroxyl group and represented by the following general formula (2A) to react with catalyst and a halogen-based silane coupling agent represented by the following general formula (2B) in a solvent.



wherein  $R^{f2}$  is a linear or branched perfluoroalkyl group having 1 to 16 carbon atoms; X is a halogen atom;  $R^2$  is a hydrolytic group;  $n2$  is an integer of 1 to 50;  $m2$  is an integer of 0 to 3; 12 is an integer of 0 to 3; and  $s2$  is an integer

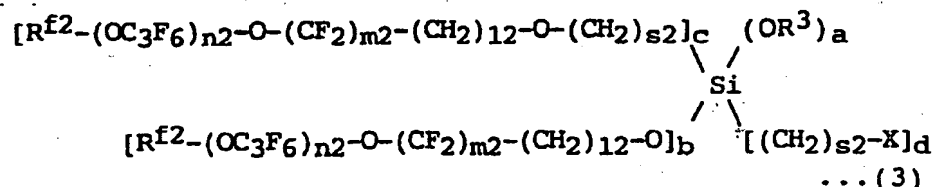
of 0 to 6.

**[0042]** This reaction can be advantageously proceeded by using 1 to 1.1 excess of NaH as a catalyst and 1 to an excess of a halogen-based silane coupling agent per 1 mol of a fluorine resin containing hydroxyl group and by performing the reaction at a temperature ranging from room temperature to the reflux temperature of the solvent for 2 to 72 hours. As for the solvent useful in this case, it is possible to employ an ordinary perfluoroaliphatic hydrocarbon having 5 to 12 carbon atoms, such as perfluorohexane, perfluoromethylcyclohexane, perfluoro-1,3-dimethylcyclohexane, etc.; an aromatic hydrocarbon polyfluoride such as bis(trifluoromethyl) benzene; or an aliphatic hydrocarbon polyfluoride.

**[0043]** The anti-smudge agent according to this invention may include, if desired, various kinds of additives in addition to the aforementioned organometallic compounds represented by the general formula (1). For example, the anti-smudge agent may contain a catalyst for promoting the hydrolysis or polycondensation of the organosilicon compounds represented by the general formula (1) or for promoting the formation of condensed ether linkage.

**[0044]** As to the catalyst that can be employed in the anti-smudge agent of this invention, it is possible to employ hydrochloric acid, nitric acid, sulfuric acid, acetic acid, hydrofluoric acid, formic acid, phosphoric acid, oxalic acid, ammonia, aluminum acetylacetonate, dibutyltin dilaurate, octylate tin compound, sodium, sodium hydride, methane sulfonate, trifluoromethane sulfonate, paratoluene sulfonate, trifluoroacetate, etc. These compounds can be employed singly or in combination of two or more kinds.

**[0045]** Another example of the organosilicon compound constituting the anti-smudge agent is an organosilicon compound represented by the following general formula (3):

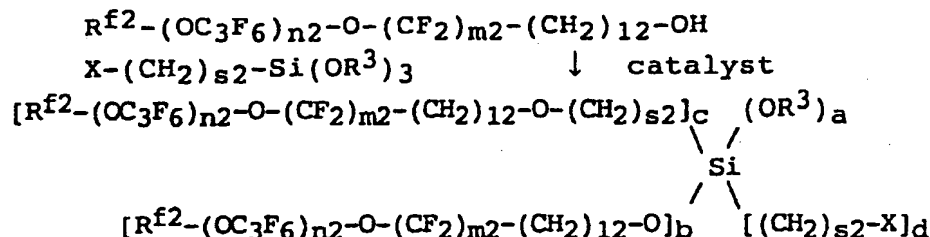


wherein  $R^2$  is a linear or branched perfluoroalkyl group having 1 to 16 carbon atoms;  $R^3$  is an alkyl group having 1 to 10 carbon atoms; X is a halogen atom; n2 is an integer of 1 to 50; m2 is an integer of 0 to 3; 12 is an integer of 0 to 3; and s2 is an integer of 1 to 6; a is an integer of 0 to 3; b is an integer of 0 to 3; c is 0 or 1; d is 0 or 1; with a proviso that these m2, 12, a, b, c and d meet the conditions of  $6 \geq m2 + 12 > 0$ ,  $a + b = 3$ , and  $c + d = 1$ .

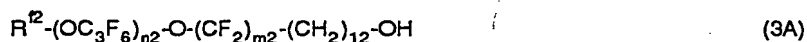
**[0046]** The number of carbon atom of the linear or branched perfluoroalkyl group to be introduced into R<sup>12</sup> is preferably in the range of from 1 to 16, more preferably in the range of from 1 to 3. Therefore, most preferable examples of R<sup>12</sup> are -CF<sub>3</sub>, -C<sub>2</sub>F<sub>5</sub>, -C<sub>3</sub>F<sub>7</sub>, etc.

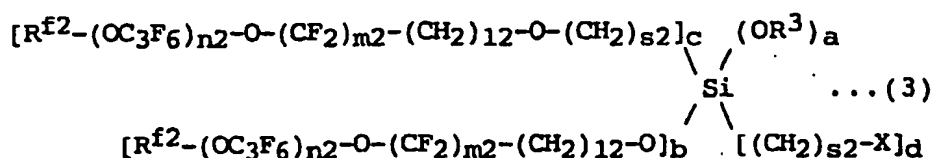
**[0047]** The number of carbon atom of the alkyl group to be introduced into R<sup>3</sup> is preferably in the range of from 1 to 10, more preferably in the range of from 1 to 3. Therefore, most preferable examples of R<sup>3</sup> are -CH<sub>3</sub>, -C<sub>2</sub>H<sub>5</sub>, -C<sub>3</sub>H<sub>7</sub>, etc.

**[0048]** The compounds represented the general formula (3) may be prepared according to the following reaction formula:



**[0049]** More specifically, an organosilicon compound represented by the following general formula (3) can be obtained by allowing a fluorine resin having hydroxyl group and represented by the following general formula (3A) to react with catalyst and a halogen-based silane coupling agent represented by the following general formula (3B) in a solvent.





[0050] This reaction can be advantageously proceeded by using 1 to 1.1 excess of NaH as a catalyst and 1 to an excess of a halogen-based silane coupling agent per 1 mol of a fluorine resin containing hydroxyl group and by performing the reaction at a temperature ranging from room temperature to the reflux temperature of the solvent for 2 to 72 hours. As for the solvent useful in this case, it is possible to employ an ordinary perfluoroaliphatic hydrocarbon having 5 to 12 carbon atoms, such as perfluorohexane, perfluoromethylcyclohexane, perfluoro-1,3-dimethylcyclohexane, etc.; an aromatic hydrocarbon polyfluoride such as bis(trifluoromethyl) benzene; or an aliphatic hydrocarbon polyfluoride.

**[0052]** The anti-smudge agent according to this invention may include, if desired, various kinds of additives in addition to the aforementioned organometallic compounds represented by the general formula (3). For example, the anti-smudge agent may contain a catalyst for promoting the hydrolysis or polycondensation of the organosilicon compounds represented by the general formula (3) or for promoting the formation of condensed ether linkage.

**[0054]** As for the treating substrate to which an anti-smudge layer is applied by making use of the anti-smudge agent of this invention, there is not any particular restriction, i.e. an ordinary optical member is applicable. For example, an inorganic substrate such as a glass board, a glass board comprising an inorganic compound layer; or an organic substrate such as a transparent plastic substrate, a transparent plastic substrate comprising an inorganic compound layer can be employed.

**[0056]** The inorganic substrate or inorganic compound layer formed of any of these inorganic compounds may be of a single layer or a multi-layer. The inorganic compound layer functions as a reflection preventive layer and can be formed by making use of any conventional method such as a wet coating method, a PVD (Physical Vapor Deposition) method and a CVD (Chemical Vapor Deposition) method. The wet coating method may be a dip coating method, a spin coating method, a flow coating method, a spray coating method, a roll coating method, a gravure coating method, etc. As for the PVD method, it is possible to employ a vacuum deposition method, a reactive vapor deposition method, an ion beam assist method, a sputtering method, an ion plating method, etc.

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terephthalate, polyethylene naphthalate, etc.), a polyamide resin (nylon-6, nylon-66, etc.), polystyrene, polyvinyl chloride, polyimide, polyvinyl alcohol, ethylene vinyl alcohol, acrylic resin, cellulose (triacetyl cellulose, diacetyl cellulose, cellophane, etc.), or a copolymer of these organic polymer may be employed. Therefore, these polymer can be employed also in this invention as a transparent plastic substrate for constituting the treating substrate.

5 [0058] It is also possible to incorporate various kinds of conventional additives such as an anti-static agent, an ultraviolet absorbent, a plasticizer, a lubricant, a colorant, an antioxidant, a flame retardant, etc. into the organic polymer constituting these organic substrates.

[0059] An inorganic compound layer may be superimposed on these organic substrates to be employed as the treating substrate in this invention. In this case, the inorganic compound layer functions as a reflection preventive layer, and can be deposited on the surface of the organic substrates in the same manner as explained above.

10 [0060] There is not any particular limitation regarding the shape of the inorganic substrate or organic substrate to be employed as the treating substrate. The transparent plastic substrate to be employed as an optical member is generally provided in the form of film or sheet, and hence, a film-like or sheet-like transparent plastic substrate can be employed in this invention also. This film-like or sheet-like transparent plastic substrate may be formed of a single layer or of a laminate layer consisting of plural organic polymer layers. Although there is not any particular limitation regarding the thickness of the transparent plastic substrate, the thickness should preferably be in the range of 0.01 to 5 mm.

15 [0061] It is also possible to interpose a hard coat layer between the transparent plastic substrate and the inorganic compound layer. The provision of this hard coat layer is effective in improving not only the hardness of the substrate surface but also the adhesivity between the transparent plastic substrate and the inorganic compound layer, since the flatness of the substrate surface can be improved by this hard coat layer. Therefore, any scratch that may be caused by the pressing of pencil, etc. on the substrate surface can be effectively prevented. Moreover, the generation of cracks of the inorganic compound layer due to the bending of the transparent plastic substrate can be effectively prevented, thus making it possible to improve the mechanical strength of the optical member.

20 [0062] There is not any particular limitation with regard to the material of the hard coat layer as long as the hard coat layer is excellent in transparency, sufficient in hardness and excellent in mechanical strength. For example, a cured resin which can be cured by the irradiation of an ionizing radiation or ultraviolet ray, or a thermosetting resin can be employed. In particular, an ultraviolet irradiation-curing type acrylic resin, an organic silicone resin and a thermosetting type polysiloxane resin are preferable for forming the hard coat layer. It is more preferable that the refractive index of these resins is the same or nearly the same as that of the transparent plastic substrate.

25 [0063] There is not any particular limitation with regard to the method of coating the hard coat layer as long as it is capable of forming a uniform layer. The thickness of the hard coat layer may be sufficient if it is 3  $\mu\text{m}$  or more for ensuring a sufficient strength. In view of the transparency, coating precision and handling thereof however, a preferable range in thickness of the hard coat layer is 5 to 7  $\mu\text{m}$ .

30 [0064] It is also possible to mix and disperse an inorganic or organic fine particle having an average particle diameter of 0.01 to 3  $\mu\text{m}$  into the hard coat layer, thereby achieving a light dispersion treatment or so-called anti-glare treatment. Although there is not any particular limitation regarding this inorganic or organic fine particle as long as the fine particle is transparent, it is more preferable to employ a material of low refractive index. Specifically, silicon oxide and magnesium fluoride is especially preferable in view of the stability and heat resistance thereof. The aforementioned light dispersion treatment can be realized also by roughening the surface of the hard coat layer.

35 [0065] The aforementioned treating substrate can be employed as a transparent substrate for the reflection preventive member in this invention. In particular, the treating substrate provided on its surface with a reflection preventive film can be treated in this invention as a transparent substrate provided with a reflection preventive film. The reflection preventive optical member according to this invention can be obtained by depositing an anti-smudge layer on the surface of the aforementioned resultant transparent substrate.

40 [0066] As for the method of forming the anti-smudge layer, there is not any particular limitation, and hence it is possible to adopt either a wet coating method or a dry coating method.

[0067] This wet coating method may be a dip coating method, a spin coating method, a flow coating method, a spray coating method, a roll coating method, a gravure coating method, etc.

45 [0068] On the other hand, the dry coating method may be for instance a vacuum deposition method, a sputtering method or a CVD method. More specifically, the vacuum deposition method may be performed by way of a resistance heating method, an electron beam method, a high frequency heating method, an ion beam method, etc. The CVD method can be performed by way of a plasma CVD method, a light CVD method, a thermal CVD method, etc.

50 [0069] Although there is not any particular limitation regarding a diluent solvent to be employed in the wet coating method, it is preferable to employ a perfluoroaliphatic hydrocarbon having 5 to 12 carbon atoms such as perfluorohexane, perfluoromethylcyclohexane, perfluoro-1,3-dimethylcyclohexane; an aromatic hydrocarbon polyfluoride such as bis(trifluoromethyl) benzene; and an aliphatic hydrocarbon polyfluoride. These solvents may be employed singly or as a mixture containing two or more kinds of these solvents.

55 [0070] The employment of this wet coating method is preferable where the treating substrate is of complicated con-



figuration or large in surface area.

[0071] On the other hand, the employment of dry coating method which can be dispensed with the employment of a diluent solvent is preferable in view of the working environment at the occasion of forming an anti-smudge layer or in view of controlling the film thickness of an anti-smudge layer. In that respect, the employment of a vacuum deposition method is most preferable. It is possible according to a dry coating method to precisely control the film thickness of the anti-smudge layer to the order of angstrom which the prior art has failed to achieve so far. As a result, it is now possible to provide an optical member having a desired anti-smudge layer. Further, it is now possible to easily provide an optical member having a reflection preventive film with an anti-smudge property without causing any change in interference color of the reflection preventive film where the color is difficult to set.

[0072] By the way, in the case where an anti-smudge layer is to be formed by means of a dry coating method, the film thickness to be formed changes depending on the quantity of evaporation of the anti-smudge agent. Therefore, if the film thickness of an anti-smudge layer is to be precisely controlled, the quantity of the evaporation of the anti-smudge agent should preferably be controlled precisely.

[0073] After finishing the deposition of an anti-smudge layer on a treating substrate by means of a dry coating method or a wet coating method, the anti-smudge layer may be subjected, if desired, to heating, humidification, light irradiation, electron beam irradiation, etc.

[0074] In particular, it is also preferable to form an anti-smudge layer by means of vacuum deposition method using a porous molded body.

[0075] Namely, this vacuum deposition method comprises the steps of; impregnating a porous molded body with an anti-smudge agent of this invention; and heating the porous molded body impregnated with the anti-smudge agent in vacuum so as to evaporate the anti-smudge agent and to deposit an anti-smudge layer on a transparent substrate covered in advance with a reflection preventive film.

[0076] As for the components constituting the porous molded body (pellet) to be employed in this method, it is possible to employ  $\text{SiO}_2$ ,  $\text{TiO}_2$ ,  $\text{ZrO}_2$ ,  $\text{MgO}$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{CaSO}_4$ , Cu, Fe, Al, stainless steel, carbon or a mixture thereof. Generally, these materials should be pulverized to have a particle diameter in the range of 5 to 20  $\mu\text{m}$ . These pulverized materials are then pelletized and sintered according to the conventional method thereby obtaining a porous molded body to be employed in this invention.

[0077] The sintering step of the pellet is generally performed at a temperature of 700 to 1,400°C for 1 to 10 hours. Depending on the particle diameter of the primary particle, density and sintering conditions, a porous molded body having a varied degree of porosity ranging from 30 to 60% can be obtained.

[0078] The anti-smudge agent of this invention is dissolved in a solvent such as perfluorohexane to obtain a solution, into which the porous molded body (pellet) is dipped so as to completely saturate the porous molded body with the anti-smudge agent. Thereafter, the solvent is allowed to evaporate, thus obtaining a porous molded body impregnated with the anti-smudge agent of this invention.

[0079] Then, by making use of this porous molded body impregnated with the anti-smudge agent, an anti-smudge layer is deposited on the surface of a treating substrate by means of a vacuum deposition method. As for the heating method for heating the porous molded body thereby to evaporate the anti-smudge agent, a resistance heating method, an electron beam heating method, ion beam heating method, high frequency heating method or a light heating method can be effectively employed.

[0080] After finishing the deposition of an anti-smudge layer on the treating substrate, the resultant substrate may be subjected, if desired, to heating, humidification, light irradiation, electron beam irradiation.

[0081] As explained above, if an anti-smudge layer is to be formed by means of a dry coating method, the quantity of the evaporation of the anti-smudge agent should preferably be controlled precisely in order to precisely control the film thickness of the anti-smudge layer. When the aforementioned porous molded body is employed, the evaporation quantity of the anti-smudge agent is determined by the impregnated quantity of the anti-smudge agent in the porous molded body and by the heating conditions at the occasion of forming the anti-smudge layer, and hence the employment of the porous molded body is advantageous in terms of controlling the film thickness. The employment of the porous molded body for forming an anti-smudge layer by means of a vacuum deposition method is also advantageous in the respects that preventing leaked anti-smudge agent due to heating from contaminate the inner portion of the apparatus and in the respects that decreasing the splash phenomenon of the anti-smudge agent.

[0082] Although there is not any particular limitation with respect to the film thickness of the anti-smudge layer to be formed using an anti-smudge agent of this invention, the film thickness should preferably be in the range of 10 to 500 angstroms in view of anti-smudge property, abrasion resistance and optical performance of an optical member.

[0083] FIG. 1 shows a cross-sectional view of one example of the reflection preventive optical member according to this invention.

[0084] Referring to FIG. 1, a reflection preventive film 2 is deposited on one of the surfaces of a transparent substrate 1. Further, an anti-smudge layer 3 is formed on this reflection preventive film 2 according to the method of this invention.

[0085] This reflection preventive optical member according to this invention is then adhered to a functional optical

member such as a polarizing plate by a suitable bonding method such as lamination thereby to obtain an optical functional member of this invention exhibiting a reflection preventive function.

[0086] FIG. 2 shows a cross-sectional view of one example of the optical functional member according to this invention.

5 [0087] Referring to FIG. 2, a reflection preventive film 2 is deposited on one of the surfaces of a transparent substrate 1. Further, an anti-smudge layer 3 is formed on this reflection preventive film 2 according to the method of this invention. Further, a polarizing plate 4 acting as a functional optical member is formed on the other surface of the transparent substrate 1.

10 [0088] This reflection preventive optical member or an optical functional member laminated with this reflection preventive optical member is then adhered, by making use of an adhesive or a pressure-sensitive adhesive, onto the front face plate (a glass plate, a plastic plate or a polarizing plate) of various display devices (a liquid crystal display, a CRT display, a projection television, a plasma display, an EL display, etc.) thereby to obtain a display device of this invention.

15 [0089] FIG. 3 shows a schematical view of one example of the display device according to this invention. In the liquid crystal display shown in FIG. 3A, a TFT substrate 10a provided on one surface thereof with an alignment film 11a is disposed to face a glass substrate 10b provided on one surface thereof with an alignment film 11b with a spacer 13 being interposed therebetween. Additionally, a liquid crystal layer 12 is kept sustained between the TFT substrate 10a and the glass substrate 10b. Further, the TFT substrate 10a is provided with a display electrode 14, and the glass substrate 10b is provided with a color filter 15 and a counter electrode 16. Further, a polarizing plate 17 is formed on the other surface of each of the TFT substrate 10a and the glass substrate 10b.

20 [0090] An optical functional member 18 provided with an anti-smudge layer which has been formed using an anti-smudge agent of this invention is formed on the surface of the polarizing plate 17 formed on the TFT substrate 10a.

[0091] FIG. 3B shows a schematical view of a portion of a CRT display. As shown in FIG. 3B, an optical functional member 22 provided with an anti-smudge layer which has been formed using an anti-smudge agent of this invention is formed on the surface of a panel 21.

25 [0092] As shown in these FIGS. 3A and 3B, an anti-smudge layer is formed on the surface of an optical functional member so as to prevent a smudge such as fingerprints, skin fat, sweat, cosmetics, etc. from adhering onto the surface of an optical functional member. This anti-smudge layer is excellent in durability and capable of easily wiping off a smudge even if the smudge is adhered onto the surface of the anti-smudge layer. Therefore, it is possible to obtain a display device which is excellent in image recognizability.

30 [0093] Next, this invention will be further explained with reference to specific examples, which are not intended to limit this invention.

[0094] First of all, a reflection preventive film-attached TAC (triacyetyl cellulose) film was prepared as follows.

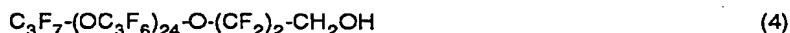
35 [0095] A monofunctional acrylic resin was coated on the surface of a TAC film (thickness: 80  $\mu\text{m}$ ) by means of a microgravure method. Then, the light from a metalhalide lamp (120W) was irradiated to the coated layer from a distance of 20 cm for 10 seconds, thereby forming a hard coat layer. Thereafter, a first  $\text{TiO}_2$  film, a first  $\text{SiO}_2$  film, a second  $\text{TiO}_2$  film and a second  $\text{SiO}_2$  film successively deposited on the hard coat layer by means of a plasma assist deposition method, thereby obtaining a reflection preventive layer.

40 [0096] The refractive index "n", geometrical thickness "d" and optical film thickness "nd" of each layer were set to: the TAC film ( $n=1.49$ ), the hard coat layer ( $n=1.51$ ,  $d=\text{about } 5 \mu\text{m}$ ), the first  $\text{TiO}_2$  film ( $n=2.30$ ,  $nd=60 \text{ nm}$ ), the first  $\text{SiO}_2$  film ( $n=1.46$ ,  $nd=40 \text{ nm}$ ), the second  $\text{TiO}_2$  film ( $n=2.30$ ,  $nd=110 \text{ nm}$ ) and the second  $\text{SiO}_2$  film ( $n=1.46$ ,  $nd=140 \text{ nm}$ ). The optical film thickness was watched by making use of an optical film thickness monitor, and when an aimed quantity of light was reached, the operation of forming the film was stopped thereby to obtain a predetermined optical film thickness.

45 [0097] In the following Examples and Comparative Examples, the reflection preventive film-attached TAC films obtained in this manner were employed as a treating substrate or as a transparent substrate for forming an anti-smudge layer.

#### Example 1

50 [0098] First of all, 1.70g of a fluorine resin having hydroxyl group and represented by the following chemical formula (4) was dissolved in 7.13g of bis(trifluoromethyl) benzene to prepare a solution. The resultant solution was then placed in a four-necked flask provided with a stirrer, a dropping funnel, a reflux condenser and a thermometer.



55 [0099] Then, 0.0018g of dibutyltin dilaurate (catalyst) was added to the solution in the flask, and the resultant mixture

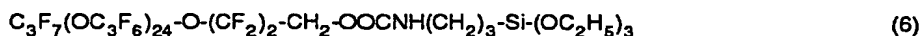
was then kept at a temperature of 100°C with sufficient stirring. Thereafter, 0.13g of isocyanate propyltriethoxy silane represented by the following chemical formula (5) was dropped from the dropping funnel into the flask.



[0100] After finishing the dropping of the silane compound, the mixture was allowed to react while being heated at reflux for 10 hours, thereby obtaining an organosilicon compound.

[0101] Then, this organosilicon compound was subjected to an FT-IR analysis to obtain an FT-IR chart as shown in FIG. 4.

[0102] As a result, the compound obtained was confirmed as being an organosilicon compound represented by the following chemical formula (6).



[0103] The organosilicon compound thus obtained was diluted with perfluorohexane to a 0.1 wt% solution thereby obtaining an anti-smudge agent of this invention.

[0104] Then, the reflection preventive film-attached TAC film prepared as mentioned above was employed as a treating substrate, and the anti-smudge agent obtained in this example was coated on the surface of the treating substrate by means of a dip coating method. The coated substrate was then heat-dried at a temperature of 60°C for one minute thereby to form an anti-smudge layer, thereby obtaining a reflection preventive optical member of this example.

[0105] FIG. 5 shows a cross-sectional view of the reflection preventive optical member obtained in this manner. As shown in FIG. 5, the reflection preventive optical member according to this example was constructed such that a hard coat layer 5 was formed on a transparent substrate 1, and a reflection preventive film 2 formed of a laminate structure consisting of a first TiO<sub>2</sub> film 2a, a first SiO<sub>2</sub> film 2b, a second TiO<sub>2</sub> film 2c and a second SiO<sub>2</sub> film 2d was deposited on the surface of hard coat layer 5. The uppermost layer of this optical member was constituted by an anti-smudge layer 3 formed from the anti-smudge agent of this invention.

#### Comparative Example 1

[0106] CF<sub>3</sub>(CF<sub>2</sub>)<sub>7</sub>(CH<sub>2</sub>)<sub>2</sub>Si(NH)<sub>3/2</sub> (KP801M; Shin-etsu Chemical Industries Ltd.) was diluted with perfluorohexane to a 0.1 wt% solution thereby obtaining an anti-smudge agent.

[0107] Then, by making use of this anti-smudge agent, an anti-smudge layer was formed on the surface of a treating substrate in the same manner as illustrated in Example 1.

#### Example 2

[0108] The organosilicon compound obtained in Example 1 and represented by the chemical formula (6) was diluted with perfluorohexane to a 20 wt% solution thereby obtaining an anti-smudge agent of this example.

[0109] Then, the reflection preventive film-attached TAC film prepared as mentioned above was employed as a treating substrate, and the anti-smudge agent obtained in this example was deposited on the surface of the treating substrate by means of a vacuum deposition method (resistance heating method), thereby forming an anti-smudge layer.

[0110] In the practice of this vacuum deposition method, 5 mg (solid matter) of the anti-smudge agent was placed inside a boat, and the interior of the deposition apparatus was evacuated to a vacuum degree of 5 × 10<sup>-5</sup> torr or less. Thereafter, the boat was heated to 400°C so as to evaporate the anti-smudge agent. Comparative Example 2

[0111] The same procedures as those of Example 2 were repeated except that CF<sub>3</sub>(CF<sub>2</sub>)<sub>7</sub>(CH<sub>2</sub>)<sub>2</sub>Si(NH)<sub>3/2</sub> (KP801M; Shin-etsu Chemical Industries Ltd.) was employed as an anti-smudge agent, thereby forming an anti-smudge layer on the surface of a treating substrate.

#### Example 3

##### (1) Manufacture of porous molded body

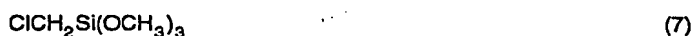
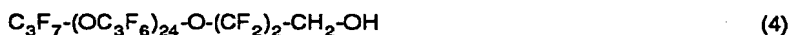
[0112] A pellet having a diameter of 6 mm and a height of 4 mm was manufactured by means of hydraulic press and by making use of Al<sub>2</sub>O<sub>3</sub> powder having a particle diameter of 1 to 10 μm as a raw material. Then, this pellet was sintered

at a temperature of 1,200°C for 14 hours. The resultant sintered substance (pellet) was found as having a porosity of about 40%.

## (2) Synthesis of an anti-smudge agent

[0113] First of all, sodium hydride oil suspension was placed in a 100 mL two-necked eggplant type flask provided with a dropping funnel and a reflux condenser thereby to perform nitrogen displacement. After this displacement, the washing with n-hexane under nitrogen gas atmosphere was repeated four times, and then n-hexane was distilled in vacuum thereby to obtain sodium hydride (0.011 mol).

[0114] 0.01 mol of a fluorine resin having hydroxyl group and represented by the following chemical formula (4) was dissolved in 50g of bis(trifluoromethyl) benzene to prepare a solution. The resultant solution was then gently dropped under ice-cooling into the flask at a rate of one drop per second. After finishing this dropping operation, the reaction mixture was taken out of the ice bath and stirred for 10 hours at room temperature. 0.1 mol of chloromethyltrimethoxy silane represented by the following formula (7) was added to sodium alkoxide and the resultant mixture was stirred for two hours at room temperature. Thereafter, the reaction mixture was heated at reflux for 72 hours while maintaining the heating temperature at 90°C. After finishing the reflux, unreacted sodium hydride and sodium chloride were filtered out under vacuum, and filtrate was water-washed sufficiently. Further, bis(trifluoromethyl) benzene and an excessive chloromethyltrimethoxy silane were distilled in vacuum thereby to obtain an organosilicon compound represented by the following formula (8).



[0115] Then, 5g of the organosilicon compound represented by the aforementioned formula (8) was diluted with perfluorohexane to a 10 wt% solution thereby obtaining an anti-smudge agent. Then, the aforementioned porous molded body (pellet) was dipped into the anti-smudge agent solution so as to completely saturate the porous molded body with the anti-smudge agent. Subsequently, the porous molded body was taken out of the anti-smudge agent solution, and the solvent was allowed to evaporate. As a result, the porous molded body (pellet) was found as containing about 2 wt% of the anti-smudge agent.

## (3) Formation of an anti-smudge layer

[0116] The pellet was mounted on a molybdenum boat, and the anti-smudge agent was allowed to deposit on the aforementioned reflection preventive film-attached TAC film by means of a vacuum deposition method (resistance heating method) thereby to form an anti-smudge layer. In this case, the interior of the deposition apparatus was evacuated to a vacuum degree of  $5 \times 10^{-5}$  torr or less. Thereafter, the boat was heated to 400°C so as to evaporate the anti-smudge agent.

## Comparative Example 3

[0117] The same procedures as those of Example 3 were repeated except that  $\text{CF}_3(\text{CF}_2)_7(\text{CH}_2)_2\text{Si}(\text{NH})_{3/2}$  (KP801M; Shin-etsu Chemical Industries Ltd.) was employed as an anti-smudge agent, thereby forming an anti-smudge layer on the surface of the reflection preventive film-attached TAC film.

## Example 4

### (1) Manufacture of porous molded body

[0118] A pellet having a porosity of about 40% was prepared in the same procedures as those of Example 3.

## (2) Synthesis of an anti-smudge agent

[0119] First of all, sodium hydride oil suspension was placed in a 100 mL two-necked eggplant type flask provided with a dropping funnel and a reflux condenser thereby to perform nitrogen displacement. After this displacement, the washing with n-hexane under nitrogen gas atmosphere was repeated four times, and then n-hexane was distilled in vacuum thereby to obtain sodium hydride (0.011 mol).

[0120] 0.01 mol of a fluorine resin having hydroxyl group and represented by the following chemical formula (4) was dissolved in 50g of bis(trifluoromethyl) benzene to prepare a solution. The resultant solution was then gently dropped under ice-cooling into the flask at a rate of one drop per second. After finishing this dropping operation, the reaction mixture was taken out of the ice bath and stirred for 10 hours at room temperature. 0.03 mol of chloromethyltrimethoxy silane represented by the following formula (7) was added to sodium alkoxide and the resultant mixture was stirred for two hours at room temperature. Thereafter, the reaction mixture was heated at reflux for 48 hours while maintaining the heating temperature at 100°C. After finishing the reflux, unreacted sodium hydride and sodium chloride were filtered out under vacuum, and filtrate was water-washed sufficiently. Further, bis(trifluoromethyl) benzene and an excessive chloromethyltrimethoxy silane were distilled in vacuum thereby to obtain an organosilicon compound represented by the following formula (9).



[0121] Then, 5g of the organosilicon compound represented by the aforementioned formula (9) was diluted with perfluorohexane to a 10 wt% solution thereby obtaining an anti-smudge agent. Then, the aforementioned porous molded body (pellet) was dipped into the anti-smudge agent solution so as to completely saturate the porous molded body with the anti-smudge agent. Subsequently, the porous molded body was taken out of the anti-smudge agent solution, and the solvent was allowed to evaporate. As a result, the porous molded body (pellet) was found as containing about 2 wt% of the anti-smudge agent.

## (3) Formation of an anti-smudge layer

[0122] By making use of the same procedures as those of Example 3, an anti-smudge layer was formed on the surface of the reflection preventive film-attached TAC film.

[0123] The anti-smudge layers obtained in the above Examples and Comparative Examples were employed as samples for evaluating the following various properties. The results obtained are summarized in the following Table 1.

(a) The measurement of contact angle: By making use of a contact angle gauge (CA-X: Kyowa Kaimen Kagaku Co. Ltd.), a droplet having a diameter of 1.8 mm was prepared at tip end of needle under a dry condition (20°C, 65%RH). Then, this droplet was contacted with the surface of a sample (solid body) so as to form a droplet on the surface of the sample. This contact angle is an angle formed between the surface of a solid body and the tangent line to the surface of the liquid at the contact point between the solid and the liquid. Therefore, the contact angle was defined as being an angle as measured from the contacting bottom of the liquid. As for the liquid, a distilled water and n-hexadecane were employed.

(b) The measurement of falling angle: By making use of a dropping angle gauge (CA-X: Kyowa Kaimen Kagaku Co. Ltd.), a droplet having a diameter of 3.0 mm was prepared at tip end of needle under a dry condition (20°C, 65%RH). Then, this droplet was contacted with the surface of a horizontal sample (solid body) so as to form a droplet on the surface of a horizontal sample. When this solid sample was gradually inclined, the droplet was caused to gradually deform, and when the sample was inclined to a certain angle, the droplet was caused to run downward. Therefore, this final inclined angle was defined as a falling angle. As for the liquid, a distilled water and n-hexadecane were employed.

(c) The adhesivity of an oil pen: By making use of an oil pen (a marker ink: lean type No.500), a straight line 1 cm in length was written on the surface of a sample so as to determine with naked eyes the easiness of adhesion and the stand-out degree of written portion. The criterion in this case was defined as follows.

- O: The written portion by the oil pen was spherically repelled;
- X: The written portion by the oil pen was not repelled and the straight line could be written;

5 (d) Easiness in wiping off of an oil ink: The oil ink that was adhered on the surface of a sample was wiped off using a cellulose nonwoven fabric (Pencot M-3; Asahi Kasei Co., Ltd.) so as to determine with naked eyes the easiness of removing it. The criterion in this case was defined as follows.

- O: The oil ink could be completely wiped off;
- Δ: It was possible to recognize a trace of oil ink after the wipe-off;
- 10 X: It was impossible to wipe off the oil ink;

15 (e) The adhesivity of finger prints: A finger was pressed on the surface of a sample for several second thereby to adhere a fingerprint on the surface of a sample so as to determine with naked eyes the easiness of adhesion and the stand-out degree of finger print. The criterion in this case was defined as follows.

- O: The adhesion of finger print was minimal and the trace of the finger print did not stand out;
- X: The adhesion of finger print could be recognized;

20 (f) Easiness in wiping off of finger prints:  
A finger print that was adhered on the surface of a sample was wiped off using a cellulose nonwoven fabric (Pencot M-3; Asahi Kasei Co., Ltd.) so as to determine with naked eyes the easiness of removing it. The criterion in this case was defined as follows.

- O: The finger print could be completely wiped off;
- 25 Δ: It was possible to recognize a trace of finger print after the wipe-off;
- X: It was impossible to wipe off the finger print;

30 (g) Abrasion resistance: The surface of a sample was rubbed 100 times while applying a load of 500gf by making use of a cellulose nonwoven fabric (Pencot M-3; Asahi Kasei Co., Ltd.), after which the aforementioned evaluations of properties (a) to (f) were performed again.

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Table 1

Example 1 Example 2 Example 3 Example 4

Contact angle (°)

Distilled water

110.5	113.5	113.0	113.3
(105.4)	(110.3)	(112.1)	(111.5)

n-hexadecane

72.5	73.2	71.8	72.8
(71.4)	(72.7)	(71.1)	(72.1)

Falling angle (°)

Distilled water

7.2	7.0	8.8	8.5
(7.3)	(7.1)	(9.9)	(9.8)

n-hexadecane

6.5	6.1	8.5	7.9
(7.0)	(6.5)	(8.9)	(8.9)

Oil ink

Adhesivity

○	○	○	○
(○)	(○)	(○)	(○)

Wipe-off property

○	○	○	○
(○)	(○)	(○)	(○)

Fingerprint

Adhesivity

○	○	○	○
(○)	(○)	(○)	(○)

Wipe-off property

○	○	○	○
(○)	(○)	(○)	(○)

Continued

Table 1

	Comparative Example 1	Comparative Example 2	Comparative Example 3
Contact angle (°)			
Distilled water	111.8 (110.4)	112.7 (110.1)	112.8 (111.1)
n-hexadecane	70.5 (70.4)	70.8 (70.7)	71.5 (71.2)
Falling angle (°)			
Distilled water	32.2 (34.0)	31.2 (33.1)	33.3 (33.0)
n-hexadecane	20.7 (22.8)	21.7 (23.9)	20.8 (22.9)
Oil ink			
Adhesivity	x (x)	Δ (Δ)	x (x)
Wipe-off property	Δ (Δ)	Δ (Δ)	Δ (Δ)
Fingerprint			
Adhesivity	x (x)	x (x)	x (x)
Wipe-off property	Δ (Δ)	Δ (Δ)	Δ (Δ)



[0124] The values in the parenthesis denote the results after the abrasion resistance test.

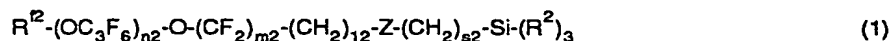
[0125] It will be seen from Table 1 that the anti-smudge layers according to Examples 1 to 4 were less wettable to distilled water and n-hexane and less adhesive to finger print and oily soil as compared with Comparative Examples 1 to 3. Moreover, even if a smudge was adhered to the anti-smudge layers according to Examples 1 to 4, the smudge could be easily removed, thus indicating an excellent anti-smudge property. Further, as shown in Table 1, the anti-smudge property of the samples of these Examples were found durable after the abrasion resistance test, thus indicating excellent durability of their anti-smudge property.

[0126] The anti-smudge layer formed on the surface of an optical member by making use of an anti-smudge agent of this invention is capable of more effectively preventing smudges formed of fingerprints, skin fat, sweat, cosmetics, etc. from being adhered onto the surface thereof without deteriorating the optical properties of the optical member as compared with the anti-smudge layer formed by making use of the conventional anti-smudge agents, and even if a smudge is adhered to the anti-smudge layer, the smudge can be easily removed, and at the same time, the durability of anti-smudge property of the anti-smudge layer of this invention is excellent.

[0127] Additionally, when an anti-smudge layer is to be formed by means of a wet coating method using an anti-smudge agent of this invention, it becomes possible to apply it to a treating substrate of complicated configuration and of large surface area, which the conventional dry coating method has failed to apply. On the other hand, when an anti-smudge layer is to be formed by means of a dry coating method using an anti-smudge agent of this invention, the employment of a diluent solvent for the formation of an anti-smudge layer can be dispensed with, and at the same time, it becomes possible to precisely control the film thickness of the anti-smudge layer. In particular, when a vacuum deposition method is employed by making use of a porous molded body for the formation of an anti-smudge layer, the controlling of film thickness would become more easy, thus being advantageous in the formation of the anti-smudge layer.

[0128] Since the optical functional member of this invention that can be obtained by the lamination of an optical member or a reflection preventive optical member of this invention with a functional optical member such as a polarizing plate is provided on the surface thereof with an anti-smudge layer which is excellent in anti-smudge property as well as in durability, it is possible to obtain a display device which is excellent in image recognizability, if this optical functional member is attached to the front face of display screen of various display devices (such as a liquid crystal display, a CRT display, a projection television, a plasma display, an EL display, etc.).

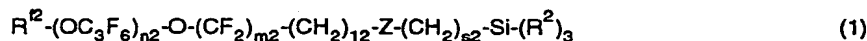
[0129] An anti-smudge agent characterized by containing an organosilicon compound represented by the following general formula (1):



wherein  $R^2$  is a linear or branched perfluoroalkyl group having 1 to 16 carbon atoms;  $R^2$  is a hydrolytic group; Z is -OOCNH- or -O-;  $n2$  is an integer of 1 to 50;  $m2$  is an integer of 0 to 3; 12 is an integer of 0 to 3; and  $s2$  is an integer of 1 to 6, with a proviso that these  $m2$  and 12 meet the condition of  $6 \geq m2 + 12 > 0$ .

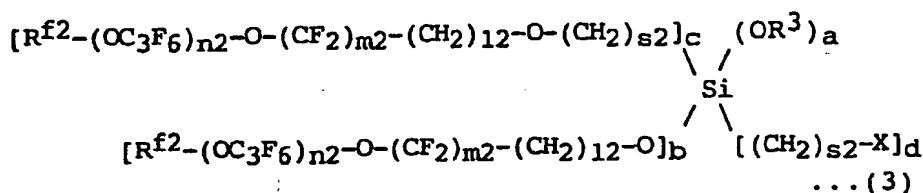
#### Claims

1. An anti-smudge agent characterized by containing an organosilicon compound represented by the following general formula (1):



wherein  $R^2$  is a linear or branched perfluoroalkyl group having 1 to 16 carbon atoms;  $R^2$  is a hydrolytic group; Z is -OOCNH- or -O-;  $n2$  is an integer of 1 to 50;  $m2$  is an integer of 0 to 3; 12 is an integer of 0 to 3; and  $s2$  is an integer of 1 to 6, with a proviso that these  $m2$  and 12 meet the condition of  $6 \geq m2 + 12 > 0$ .

2. The anti-smudge agent according to claim 1, characterized in that said  $R^2$  in said general formula (1) is alkoxy group having 1 to 10 carbon atoms.
3. An anti-smudge agent characterized by containing an organosilicon compound represented by the following general formula (3):



wherein R<sup>2</sup> is a linear or branched perfluoroalkyl group having 1 to 16 carbon atoms; R<sup>3</sup> is an alkyl group having 1 to 10 carbon atoms; X is a halogen atom; n<sub>2</sub> is an integer of 1 to 50; m<sub>2</sub> is an integer of 0 to 3; 1<sub>2</sub> is an integer of 0 to 3; and s<sub>2</sub> is an integer of 1 to 6; a is an integer of 0 to 3; b is an integer of 0 to 3; c is 0 or 1; d is 0 or 1; with a proviso that these m<sub>2</sub>, 1<sub>2</sub>, a, b, c and d meet the conditions of 6 ≥ m<sub>2</sub> + 1<sub>2</sub> > 0, a + b = 3, and c + d = 1.

- 15 4. The anti-smudge agent according to claim 3, characterized in that said organosilicon compound is formed of a first organosilicon compound represented by said general formula (3) and a second organosilicon compound represented by said general formula (3), and said first organosilicon compound and said second organosilicon compound are different from each other in the components "a", "b", "c" and "d" in said general formula (3).
- 20 5. A method of forming an anti-smudge layer characterized by comprising a step of depositing the anti-smudge agent of claim 1 on a treating substrate by means of a wet coating method.
- 25 6. The method of forming an anti-smudge layer according to claim 5, characterized in that further comprising a step of subjecting said anti-smudge layer that has been formed on the treating substrate by making use of said anti-smudge agent to heating, humidification, light irradiation, or electron beam irradiation.
- 30 7. A method of forming an anti-smudge layer characterized by comprising a step of depositing the anti-smudge agent of claim 3 on a treating substrate by means of a wet coating method.
- 35 8. The method of forming an anti-smudge layer according to claim 7, characterized in that further comprising a step of subjecting said anti-smudge layer that has been formed on the treating substrate by making use of said anti-smudge agent to heating, humidification, light irradiation, or electron beam irradiation.
- 40 9. A method of forming an anti-smudge layer characterized by comprising a step of depositing the anti-smudge agent of claim 1 on a treating substrate by means of a dry coating method.
- 45 10. The method of forming an anti-smudge layer according to claim 9, characterized in that further comprising a step of subjecting said anti-smudge layer that has been formed on the treating substrate by making use of said anti-smudge agent to heating, humidification, light irradiation, or electron beam irradiation.
- 50 11. A method of forming an anti-smudge layer characterized by comprising a step of depositing the anti-smudge agent of claim 3 on a treating substrate by means of a dry coating method.
- 55 12. The method of forming an anti-smudge layer according to claim 11, characterized in that further comprising a step of subjecting said anti-smudge layer that has been formed on the treating substrate by making use of said anti-smudge agent to heating, humidification, light irradiation, or electron beam irradiation.
13. A method of forming an anti-smudge layer characterized by comprising the steps of;
- 50       impregnating a porous molded body with the anti-smudge agent of claim 1; and  
      heating said porous molded body impregnated with said anti-smudge agent in vacuum thereby to evaporate said anti-smudge agent and to form an anti-smudge layer on a treating substrate.
- 55 14. The method of forming an anti-smudge layer according to claim 13, characterized in that further comprising a step of subjecting said anti-smudge layer that has been formed on the treating substrate by making use of said anti-smudge agent to heating, humidification, light irradiation, or electron beam irradiation.
15. The method of forming an anti-smudge layer according to claim 13, characterized in that said porous molded body

is constituted by at least one kind of materials selected from the group consisting of  $\text{SiO}_2$ ,  $\text{TiO}_2$ ,  $\text{ZrO}_2$ ,  $\text{MgO}$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{CaSO}_4$ , Cu, Fe, Al, stainless steel and carbon.

- 5 16. The method of forming an anti-smudge layer according to claim 13, characterized in that the heating method for evaporating said anti-smudge agent impregnated in said porous molded body is at least one method selected from a resistance heating method, an electron beam heating method, ion beam heating method, high frequency heating method and a light heating method.
- 10 17. A method of forming an anti-smudge layer characterized by comprising the steps of;  
impregnating a porous molded body with the anti-smudge agent of claim 3; and  
heating said porous molded body impregnated with said anti-smudge agent in vacuum thereby to evaporate said anti-smudge agent and to form an anti-smudge layer on a treating substrate.
- 15 18. The method of forming an anti-smudge layer according to claim 17, characterized in that further comprising a step of subjecting said anti-smudge layer that has been formed on the treating substrate by making use of said anti-smudge agent to heating, humidification, light irradiation, or electron beam irradiation.
- 20 19. The method of forming an anti-smudge layer according to claim 17, characterized in that said porous molded body is constituted by at least one kind of materials selected from the group consisting of  $\text{SiO}_2$ ,  $\text{TiO}_2$ ,  $\text{ZrO}_2$ ,  $\text{MgO}$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{CaSO}_4$ , Cu, Fe, Al, stainless steel and carbon.
- 25 20. The method of forming an anti-smudge layer according to claim 17, characterized in that the heating method for evaporating said anti-smudge agent impregnated in said porous molded body is at least one method selected from a resistance heating method, an electron beam heating method, ion beam heating method, high frequency heating method and a light heating method.
- 30 21. A reflection preventive optical member comprising a transparent substrate; a reflection preventive film formed on at least one surface of said transparent substrate; and an anti-smudge layer formed on an outermost surface; characterized in that said anti-smudge layer contains the anti-smudge agent of claim 1.
- 35 22. The reflection preventive optical member according to claim 21, characterized in that said transparent substrate is an organic substrate formed of a transparent substrate or an inorganic substrate formed of a glass substrate.
- 40 23. A reflection preventive optical member comprising a transparent substrate; a reflection preventive film formed on at least one surface of said transparent substrate; and an anti-smudge layer formed on an outermost surface; characterized in that said anti-smudge layer contains the anti-smudge agent of claim 3.
- 45 24. The reflection preventive optical member according to claim 23, characterized in that said transparent substrate is an organic substrate formed of a transparent substrate or an inorganic substrate formed of a glass substrate.
- 50 25. An optical functional member characterized by comprising the reflection preventive optical member of claim 21; and a functional optical member adhered to said reflection preventive optical member.
- 55 26. The optical functional member according to claim 25, characterized in that said functional optical member is a polarizing plate.
27. An optical functional member characterized by comprising the reflection preventive optical member of claim 23; and a functional optical member adhered to said reflection preventive optical member.
28. The optical functional member according to claim 27, characterized in that said functional optical member is a polarizing plate.
29. A display device comprising a display screen having front face, on the surface of which a coat member is adhered via an adhesive, characterized in that said coat member is the optical functional member of claim 25.
30. A display device comprising a display screen having front face, on the surface of which a coat member is adhered via an adhesive, characterized in that said coat member is the optical functional member of claim 26.

31. A display device comprising a display screen having front face, on the surface of which a coat member is adhered via an adhesive, characterized in that said coat member is the optical functional member of claim 27.
- 5 32. A display device comprising a display screen having front face, on the surface of which a coat member is adhered via an adhesive, characterized in that said coat member is the optical functional member of claim 28.
33. The display device according to claim 29, characterized in that said display is a liquid crystal display, a CRT display, a projection television, a plasma display, or an EL display.
- 10 34. The display device according to claim 30, characterized in that said display is a liquid crystal display, a CRT display, a projection television, a plasma display, or an EL display.
35. The display device according to claim 31, characterized in that said display is a liquid crystal display, a CRT display, a projection television, a plasma display, or an EL display.
- 15 36. The display device according to claim 32, characterized in that said display is a liquid crystal display, a CRT display, a projection television, a plasma display, or an EL display.

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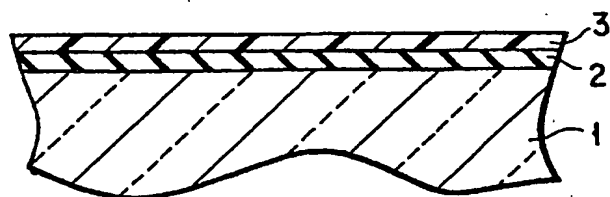


FIG. 1

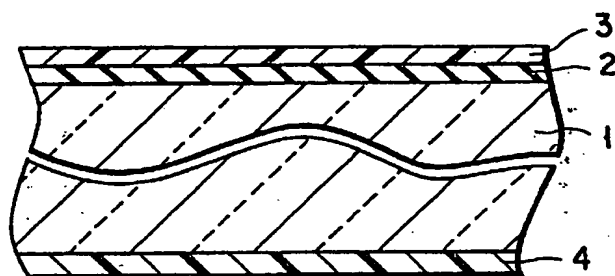


FIG. 2

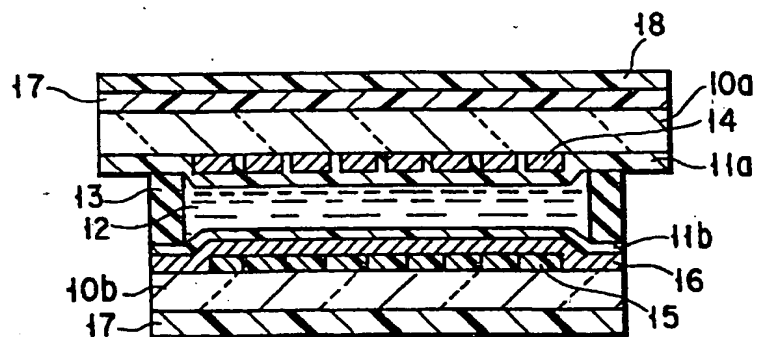


FIG. 3A

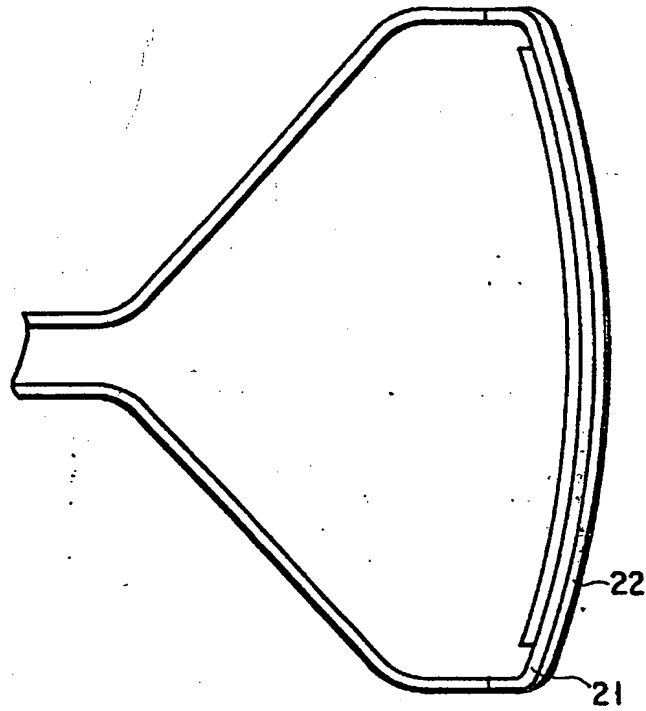


FIG. 3B

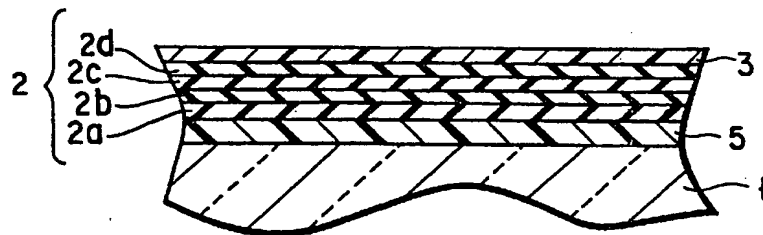


FIG. 5

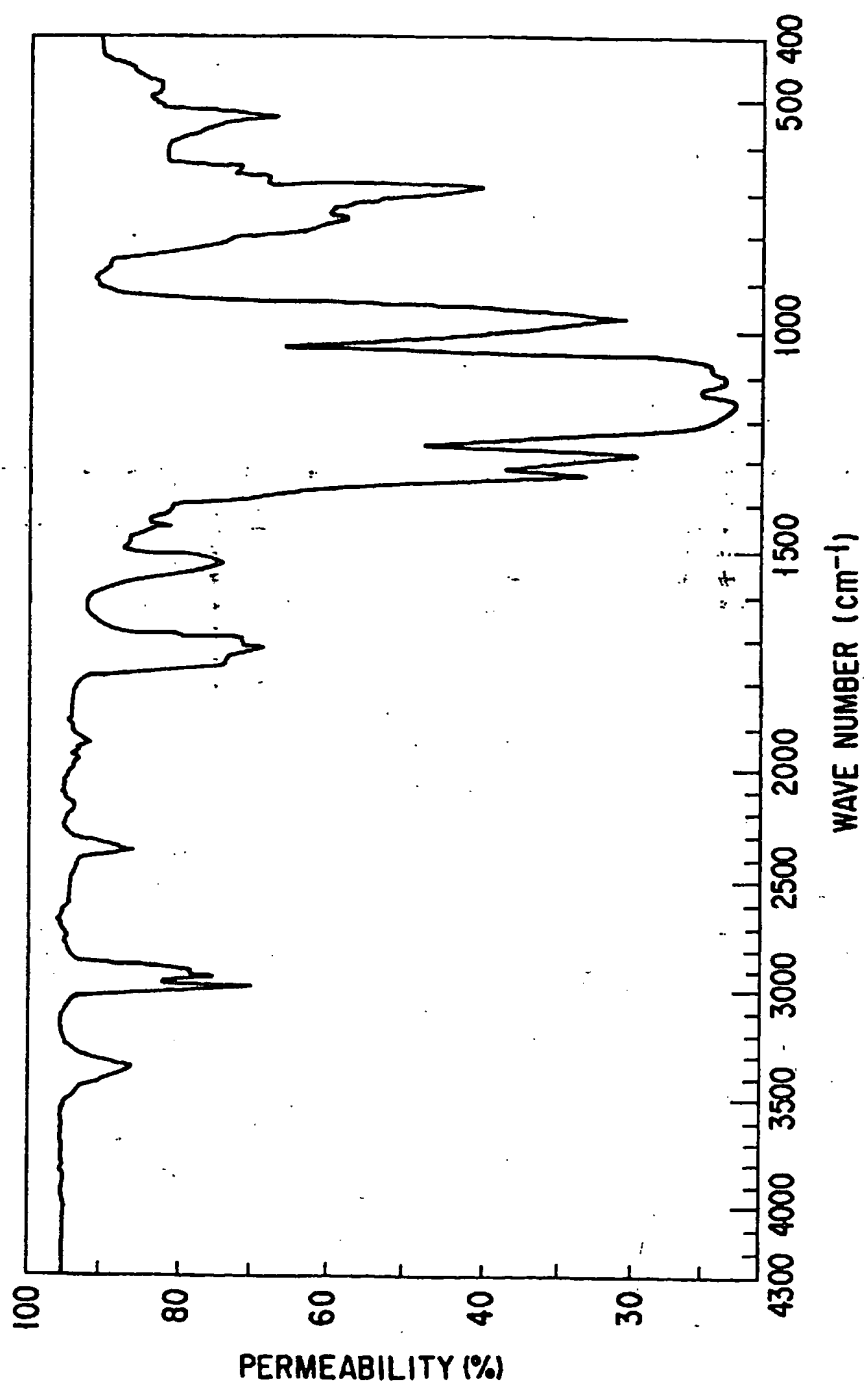
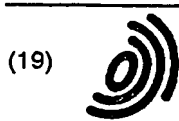


FIG. 4



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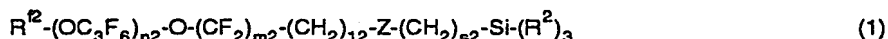
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(54) Anti-smudge agent, method of forming an anti-smudge layer, anti-reflection optical member and display device

(57) An anti-smudge agent characterized by containing an organosilicon compound represented by the following general formula (1):



wherein  $R^2$  is a linear or branched perfluoroalkyl group having 1 to 16 carbon atoms;  $R^2$  is a hydrolytic group; Z is  $-OOCNH-$  or  $-O-$ ;  $n2$  is an integer of 1 to 50;  $m2$  is an integer of 0 to 3; 12 is an integer of 0 to 3; and  $s2$  is an integer of 1 to 6, with a proviso that these  $m2$  and 12 meet the condition of  $6 \geq m2 + 12 > 0$ .

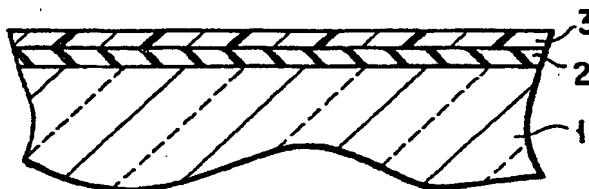


FIG. 1

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## EUROPEAN SEARCH REPORT

Application Number  
EP 99 10 1960

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)
X	US 4 687 707 A (MATSUO MASASHI ET AL) 18 August 1987 (1987-08-18) * column 2, line 10 - line 39 * * column 3, line 49 - column 4, line 25 * * column 5, line 5,12,15,17,22,25,27 * * column 6, line 11 - line 36 * * claims *	1-36	C07F7/18 G02B1/10 G02B1/11
X	DATABASE WPI Section Ch, Week 199237 Derwent Publications Ltd., London, GB; Class A26, AN 1992-305510 XP002151919 -& JP 04 213384 A (ASAHI GLASS CO LTD), 4 August 1992 (1992-08-04)	1,2	
A	* abstract * * Patent, column 3, line 1-21 *	4-36 *	
A	US 5 461 173 A (SATO SHINICHI ET AL) 24 October 1995 (1995-10-24) * column 8, structure 2-1 *	1,2	
A	US 4 944 962 A (FURUTA SHOICHIRO ET AL) 31 July 1990 (1990-07-31) * column 2, line 14 - line 54 * * column 2, line 66 - column 3, line 15 * * claims *	1-36	TECHNICAL FIELDS SEARCHED (Int.Cl.6)  C07F G02B
A	EP 0 797 111 A (SONY CORP) 24 September 1997 (1997-09-24) * page 3, line 1 - line 56 * * page 4, line 5 - line 9 * * page 7, line 40 - line 58 * * table 1 * * claims 1-4 *	1-36	
The present search report has been drawn up for all claims			
Place of search BERLIN		Date of completion of the search 3 November 2000	Examiner O'Sullivan, P
CATEGORY OF CITED DOCUMENTS X: particularly relevant if taken alone Y: particularly relevant if combined with another document of the same category A: technological background O: non-written disclosure P: intermediate document		T: theory or principle underlying the invention E: earlier patent document, but published on, or after the filing date D: document cited in the application L: document cited for other reasons &: member of the same patent family, corresponding document	

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ANNEX TO THE EUROPEAN SEARCH REPORT  
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EP 99 10 1960

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For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

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81925 München (DE)**(54) FLUOROPOLYMER, PROCESS FOR PREPARING THE SAME, AND USE THEREOF**

(57) A fluorine-containing polymer, having a number-average molecular weight of 500 to 100,000 and represented by the following general formula

$$\text{Rf}-(\text{OC}_3\text{F}_6)_a-(\text{OC}_2\text{F}_4)_b-(\text{OCF}_2)_c-\text{X}-\text{Y}-\text{Z}-\text{MP}-$$

$$\text{or} \quad \text{P}_n\text{R}_m-$$

$$\text{Rf}-(\text{OC}_3\text{F}_6)_a-(\text{OC}_2\text{F}_4)_b-(\text{OCF}_2)_c-\text{X}-\text{Y}-\text{Z}-\text{MP}_n\text{R}_m-$$

(wherein Rf is a perfluoroalkyl group; a, b and c independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of a, b and c is at least 1; X is a group represented by the formula:  $-(\text{O})_d-(\text{CF}_2)_e-(\text{CH}_2)_f-$ ; Y is a divalent polar group; Z is a group represented by the formula:  $-(\text{CH}_2)_g-$ ; M is a metal atom; R is a hydrocarbon group; P is a hydrolyzable polar group; m is an integer equal to "(valency of M) - 1"; and n is an integer of 1 to m), exhibits excellent stainproof property against oily dirt and can form on a substrate a stainproof layer having an excellent stainproof property against particularly fingerprints.

EP 0 978 524 A1

## Description

## FIELD OF THE INVENTION

- 5 [0001] The present invention relates to a fluorine-containing polymer, a process for preparing the polymer, and the use of the polymer, for example, in a stainproof substrate having a stainproof layer of said polymer on a surface thereof.

## RELATED ART

- 10 [0002] Materials such as metals, glass and resins are used as general-purpose substrates for a variety of articles such as automotive parts, OA apparatus and home electric appliances. The surfaces of these substrates are liable to get stained by the adhesion of dust particles suspended in the atmosphere of car interior, office interior, room interior and the like, and by the adhesion of oily substances which are contained in foods, oils for machines and the like and are difficult to be wiped off, and also by the adhesion of fingerprints of users. Therefore, these substrates need to be sub-  
15 jected to a stainproof treatment so that the substrate surface is rendered less adherent to dirt and the dirt once adhered can be easily removed.

- [0003] As to the stainproof treatment of glass surface, publications such as Japanese Patent Application Laid-Open No. 126244/1989 (JP-A-1-126244) disclose techniques wherein coating films are formed either by direct application of a polymeric substance such as a polydimethylsiloxane to the glass surface or by immersion of glass in a treatment  
20 agent containing such a polymeric substance. Also known is a technique wherein a chemical absorption fluorine-containing monomolecular film is formed on the glass surface (see JP-A-4-132637, etc.).

- [0004] As to the stainproof treatment of metal surface, Japanese Patent Kokoku Publication No. 53913/1995 (JP-B-7-53913) discloses a technique relating to an organic composite-plated steel plate by the steps of forming a chromate coating containing a silica sol silane coupling agent on a galvanized steel and thereafter forming a thin layer with an  
25 isocyanate-based paint composition.

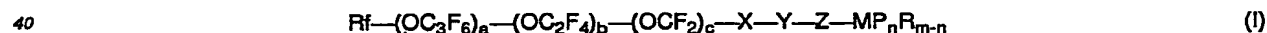
[0005] However, the above-described treatments of prior arts do not provide a sufficient stainproof against oily dirt. In addition, the prior art treatments are associated with the problem that fingerprints tend to adhere to the substrate surface which is touched with fingers and the removal therefrom of the fingerprints is difficult.

## 30 SUMMARY OF THE INVENTION

[0006] A first object of the present invention is to provide a fluorine-containing polymer which can form on a substrate a stainproof layer exhibiting an excellent stainproof property against oily dirt, particularly fingerprints, as well as a process for preparing the polymer.

- 35 [0007] A second object of the present invention is to provide a stainproof substrate which exhibits an excellent stainproof property against oily dirt, particularly against fingerprints, as well as a process for preparing the substrate.

[0008] According to the present invention, the first object can be achieved by a fluorine-containing polymer having a number-average molecular weight of 500 to 100,000 and represented by the following general formula (I):



wherein Rf is a perfluoroalkyl group;

- 45 a, b and c independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of a, b and c is at least 1;

X is a group represented by the formula:  $-(O)_d-(CF_2)_e-(CH_2)_f-$  (where d, e and f independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of e and f is at least 1, and the presence order of the bracketed repeating units is arbitrary in the formula);

Y is a divalent polar group;

- 50 Z is a group represented by the formula:  $-(CH_2)_g-$  (where g is 0 or a number equal to or greater than 1);

M is a metal atom;

R is a hydrocarbon group;

P is a hydrolyzable polar group;

m is an integer equal to "(valency of M) - 1";

- 55 n is an integer of 1 to m;

with the proviso that the presence order of the bracketed repeating units is arbitrary in the formula;  $-OC_3F_6-$  represents either  $-OCF_2CF_2CF_2-$  or  $-OCF(CF_3)CF_2-$ ; and  $-OC_2F_4-$  represents either  $-OCF_2CF_2-$  or  $-OCF(CF_3)-$ .

[0009] A preferred example of the fluorine-containing polymer (I) is a fluorine-containing polymer having a number-average molecular weight of 500 to 100,000 and represented by the following general formula (III):



wherein Rf, X, Y, Z, P and R are the same as defined above, and n is an integer of 1 to 3, with the proviso that a is a number equal to or greater than 1.

[0010] Likewise, the first object can be achieved by a fluorine-containing polymer having a number-average molecular weight of 500 to 100,000 and represented by the following general formula (II):



wherein X, Y, Z, M, P, R, a, b, c, m and n are the same as defined above.

[0011] In the general formulae (I) and (III), Rf may be any perfluoroalkyl group contained in known fluorine-containing organic polymers, and may be, for example, a linear or branched perfluoroalkyl group having 1 to 16 carbon atoms. Preferably, Rf is  $\text{CF}_3-$ ,  $\text{C}_2\text{F}_5-$  or  $\text{C}_3\text{F}_7-$ .

[0012] In the general formulae (I) and (II), a, b and c represent, respectively, the numbers of the three-type repeating units in the perfluoropolyethers constituting the main skeleton of the fluorine-containing polymer, wherein a, b and c are each 0 or a positive number equal to or greater than 1, with the proviso that  $a+b+c$  is at least 1. Preferably, a, b and c are independently selected from 0 and a number in the range of from 1 to 200. More preferably, a, b and c are each 1 to 100 when the number-average molecular weight of the fluorine-containing polymer is taken into consideration.

[0013] The presence orders of the repeating units in brackets having subscripts a, b and c are described according to specific presence orders in the general formulae (I) and (II) for the purpose of convenience. However, the linkage orders of these repeating units are arbitrary without being limited by these orders.

[0014] In the general formulae (I), (II) and (III), X is a group represented by the formula:  $-(\text{O})_d-(\text{CF}_2)_e-(\text{CH}_2)_f-$ . In this formula, d, e and f independently represent 0 or a positive number equal to or greater than 1 (e.g., 1 to 50), with the proviso that  $e+f$  is at least 1, and the presence order of the repeating units in brackets having subscripts d, e and f is not limited in the formula. Preferably, d, e and f are each 0, 1 or 2. More preferably, d is 0 or 1, e is 2, and f is 0 or 1.

[0015] In the general formulae (I), (II) and (III), Y is a divalent polar group. Examples of the divalent polar group may include  $-\text{COO}-$ ,  $-\text{OCO}-$ ,  $-\text{CONH}-$ ,  $-\text{NHCO}-$ ,  $-\text{OCH}_2\text{CH}(\text{OH})\text{CH}_2-$ ,  $-\text{CH}_2\text{CH}(\text{OH})\text{CH}_2\text{O}-$ ,  $-\text{COS}-$ ,  $-\text{SCO}-$  and  $-\text{O}-$ . Preferably, Y is  $-\text{COO}-$ ,  $-\text{CONH}-$ ,  $-\text{OCH}_2\text{CH}(\text{OH})\text{CH}_2-$  or  $-\text{CH}_2\text{CH}(\text{OH})\text{CH}_2\text{O}-$ .

[0016] In the general formulae (I), (II) and (III), Z is a group represented by the formula:  $-(\text{CH}_2)_g-$ . In this formula, g represents 0 or a positive number equal to or greater than 1 (e.g., 1 to 50). Preferably, g is 0, 1, 2 or 3.

[0017] In the general formulae (I), (II) and (III), M is a metal atom. Examples of the metal atom may include metallic elements which constitute Groups 1 to 15 (e.g., Si, Al and Ti) of the Periodic Table.

[0018] In the general formulae (I) and (II), m is equal to "(valency of M) - 1", and 0 or a positive number, while n is an integer of 1 to m. The valency of M is ordinarily 1 to 5, for example, 2 to 5, and particularly 3 to 5.

[0019] In the general formula (III), n is an integer of 1 to 3.

[0020] For example, in the case where M is silicon (Si), m is 3, and n is 1, 2 or 3. Meanwhile, in some cases, the fluorine-containing polymer is present as a mixture of polymers which are represented by the general formula (I), (II) or (III) and have different values for n. If the fluorine-containing polymer is present as a mixture of polymers as described above, n can be given as an average in the mixture.

[0021] In the general formulae (I), (II) and (III), the hydrocarbon group represented by R is preferably a hydrocarbon group containing 1 to 5 carbon atoms. Specific examples of the hydrocarbon group may include alkyl groups such as



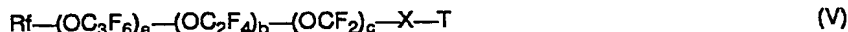
[0022] Examples of the hydrolyzable polar group represented by P may include halogen,  $-\text{OA}$ ,  $-\text{OCoA}$ ,  $-\text{O}-\text{N}=\text{C}(\text{A})_2$  (where A is an alkyl group such as  $-\text{CH}_3$ ,  $-\text{C}_2\text{H}_5$  and  $-\text{C}_3\text{H}_7$ ).

[0023] The number-average molecular weight of the fluorine-containing polymer is ordinarily 500 to 100,000. If the number-average molecular weight is less than 500, the polymer is useless because good properties of a polymer are not exhibited. On the other hand, if the number-average molecular weight is more than 100,000, the processability of the polymer is impaired. Preferably, the number-average molecular weight is 1,000 to 10,000.

[0024] The polymer (I) can be prepared by the reaction preferably between a compound represented by the following general formula (IV):



wherein Z, M, P, R, m and n are the same as defined above, and Q is a polar group, and either a compound represented by the following general formula (V):



wherein Rf, X, a, b and c are the same as defined above, and T is a polar group, or a compound represented by the following general formula (VIII):

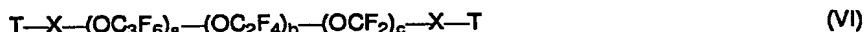


wherein Rf, X and a are the same as defined above, and T is a polar group.

[0025] Further, the polymer (II) can be synthesized by the reaction preferably between a compound represented by the following general formula (IV):



wherein Z, M, P, R, m and n are the same as defined above, and Q is a polar group, and a compound represented by the following general formula (VI):



wherein X, a, b and c are the same as defined above, and T is a polar group.

[0026] Furthermore, the polymer (III) can be prepared by the reaction preferably between a compound represented by the following general formula (VII):

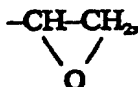


wherein Z, P and R are the same as defined above, n is an integer of 1 to 3, and Q is a polar group, and a compound represented by the following general formula (VIII):



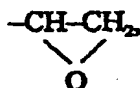
wherein Rf, X and a are the same as defined above, and T is a polar group.

[0027] Examples of the polar group represented by Q may include —OH, —NH<sub>2</sub>, —SH,



—Hal (halogen) and —COOH.

[0028] Examples of the polar group represented by T may include —OH, —COOH, —NH<sub>2</sub>,



and —CO—Hal (acid halide).

[0029] The second object of the present invention can be achieved by a stainproof substrate comprising a substrate, and a layer of the fluorine-containing polymer presented by the general formula (I), (II) or (III) on the substrate.

[0030] The polymer layer can be formed by applying a separately prepared polymer (I), (II) or (III) to the substrate surface. On the other hand, in a preferred embodiment the polymer layer (I), (II) or (III) can be formed by a process comprising forming on the substrate surface a layer of a compound represented by the general formula (IV) and reacting therewith a compound represented by the general formula (V) or (VI) or alternatively by a process comprising forming on the substrate surface a layer of a compound presented by the general formula (VII) and reacting therewith a compound represented by the general formula (VIII).

[0031] In the case where a layer of the fluorine-containing polymer is formed by a coating method, a solution, suspension liquid or dispersion liquid of the fluorine-containing polymer is coated on the substrate surface, and then is dried.

The coating method may be a conventionally known one and examples of the coating method may include spraying, spin coating, immersion, roll coating, gravure coating and curtain flow coating.

[0032] A solvent in which the polymer is dissolved, suspended or dispersed is not particularly limited. Preferred examples of the solvent may include perfluorohexane, perfluoro-1,3-dimethylcyclohexane and dichloropentafluoropropane (HCFC-225).

[0033] In the case where a layer of the fluorine-containing polymer is formed by successively applying a compound (IV) and a compound (V) to the substrate surface and reacting the two compounds, and examples of the method for coating the compounds (IV) and (V) may include spraying, spin coating, immersion, roll coating, gravure coating and curtain flow coating.

[0034] When applying the compounds (IV) and (V), these compounds are preferably dissolved, suspended or dispersed in a solvent. The solvent is not particularly limited. For example, in the case of the compound (IV), conventionally used organic solvents, such as acetone, methyl isobutyl ketone and ethanol, can be used, while, in the case of the compound (V), useful solvents include perfluorohexane, perfluoro-1,3-dimethylcyclohexane and dichloropentafluoropropane (HCFC-225).

[0035] Although the thickness of the fluorine-containing polymer layer in the stainproof substrate of the present invention is not limited, the thickness is preferably 0.001  $\mu\text{m}$  to 0.03  $\mu\text{m}$ .

[0036] The types of the substrate to be stainproof in the present invention are not particularly limited. Accordingly, examples of the substrate may include glass, resins, metals, ceramics, wood, porcelain, stone and leather.

[0037] As to the glass substrates, the substrates are not particularly limited so long as they are made from glass. Examples of the substrate may include glass for use on the surface of a variety of articles such as show windows, mirrors, water baths, windowpanes, tablewares and glass cases.

[0038] As to the resin substrates, the substrates are not particularly limited and examples thereof include articles made not only from natural resins but also from synthetic resins.

[0039] Examples of the natural resins may include cellulose and japan lacquer. Examples of the synthetic resins may include polyamide resins, polyacrylate resins, polyamideimide resins, polyvinyl acetate resins, polyvinyl chloride resins, phenol resins, urea resins, melamine resins, epoxy resins and polyester resins.

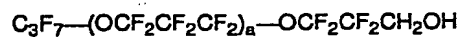
[0040] Examples of the metal substrate may include iron, zinc, lead, copper and aluminum.

[0041] Meanwhile, the compound (VII) (silane coupling agent and the like), the compound (V) and the compound (VI) are all commercialized and are easily available.

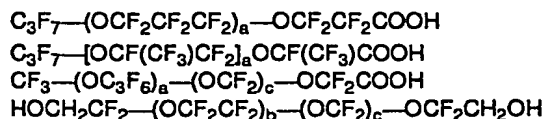
[0042] For example, the following compounds are commercially available as the compound (VII) (silane coupling agent and the like) by Toray Dow Coring Co., Ltd.

Product name	Chemical name	Structural formula	Typical properties		
			Molecular weight	Specific gravity	Refractive index
AX43-062	$\gamma$ -mercaptopropylmethyldimethoxysilane	$\text{HSCH}_2\text{CH}_2\text{CH}_2\text{Si}(\text{OCH}_3)_2$   $\text{CH}_3$	180.3	0.99	1.449
AX43-024	hydroxypropyltrimethoxysilane	$\text{HOCH}_2\text{CH}_2\text{CH}_2\text{Si}(\text{OCH}_3)_3$	180.3	0.90	1.365
SH6020	$\gamma$ -(2-aminoethyl)aminopropyltrimethoxysilane	$\text{NH}_2\text{CH}_2\text{CH}_2\text{NHCH}_2\text{CH}_2\text{CH}_2\text{Si}(\text{OCH}_3)_3$	222.1	1.03	1.444
SH6020P	$\gamma$ -(2-aminoethyl)aminopropyl-methyldimethoxysilane	$\text{NH}_2\text{CH}_2\text{CH}_2\text{NHCH}_2\text{CH}_2\text{CH}_2\text{Si}(\text{OCH}_3)_2$   $\text{CH}_3$	206.1	0.97	1.449
SH6040	$\gamma$ -glycidoxypropyltrimethoxysilane	$\text{CH}_2-\text{CH}-\text{CH}_2\text{OCH}_2\text{CH}_2\text{CH}_2\text{Si}(\text{OCH}_3)_3$   $\text{O}$	236.1	1.07	1.427
SH6076	$\gamma$ -chloropropyltrimethoxysilane	$\text{ClCH}_2\text{CH}_2\text{CH}_2\text{Si}(\text{OCH}_3)_3$	198.5	1.08	1.418
SH6062	$\gamma$ -mercaptopropyltrimethoxysilane	$\text{HSCH}_2\text{CH}_2\text{CH}_2\text{Si}(\text{OCH}_3)_3$	196.2	1.06	1.442
AX43-110	$\gamma$ -chloropropylmethyldimethoxysilane	$\text{ClCH}_2\text{CH}_2\text{CH}_2\text{Si}(\text{OCH}_3)_2$   $\text{CH}_3$	182.7	1.03	1.426

[0043] On the other hand, the following compounds are commercially available as the compounds (V) and (VI).







[0044] In addition,  $\text{C}_3\text{F}_7-(\text{OC}_3\text{F}_6)_a-\text{OCF}_2\text{CF}_2\text{COF}$  is available as one of raw materials for these compounds (JP-B No. Showa 63-43419 (43419/1988)). A compound such as  $\text{C}_3\text{F}_7-(\text{OC}_3\text{F}_6)_a-\text{OCF}_2\text{CF}_2\text{CN}$  can be derived therefrom (Ind. Eng. Chem. Res., 1987, 26, 1980). Reduction of the foregoing compound provides  $\text{C}_3\text{F}_7-(\text{OC}_3\text{F}_6)_a-\text{OCF}_2\text{CF}_2\text{CH}_2\text{NH}_2$ .

[0045] The reaction between  $\text{C}_3\text{F}_7-(\text{OC}_3\text{F}_6)_a-\text{OC}_2\text{F}_4\text{CH}_2\text{OH}$  and epichlorohydrin provides

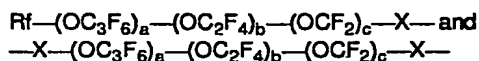


[0046] The reaction between  $\text{C}_3\text{F}_7-(\text{OC}_3\text{F}_6)_a-\text{OC}_2\text{F}_4\text{CH}_2\text{OH}$  and a metal hydride  $\text{MH}_x$  (M is a metal and x is a number of 1 to 6) (e.g., KH and NaH) provides  $\text{C}_3\text{F}_7-(\text{OC}_3\text{F}_6)_a-\text{OC}_2\text{F}_4\text{CH}_2\text{OM}$  (metal alcoholate).

[0047] A fluorine-containing compound (I) or (II) terminated with a metal coupling group can be obtained by chemically combining the perfluoropolyether (V) or (VI) with the compound (VII) through the reaction between the polar groups thereof.

[0048] Examples of the reaction between the polar groups are given below.

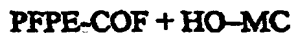
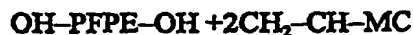
[0049] In the following, for the purpose of simplification,



in the general formulae (V) and (VI) are expressed as "PFPE". Likewise,  $\text{Z}-\text{M}-\text{P}_n\text{R}_{m-n}$  in the general formula (IV) is expressed as "MC".

[0050] Accordingly, the general formula (V) is expressed as "PFPE-T", the general formula (VI) as "T-PFPE-T", and the general formula (IV) as "MC-Q". In this case, the general formula (I) is "PFPE-Y-MC" and the general formula (II) is "MC-Y-PFPE-Y-MC".

[0051] If an ordinary polar group is selected as T and Q, the fluorine-containing polymers (I) and (II) can be obtained by, for example, the following reactions.



[0052] Since a metal coupling material is unstable to water, it is preferable that the coating film on the substrate surface is formed by a procedure in which a compound (IV) is first applied to the substrate surface and thereafter the compound (IV) is reacted with a compound (V) or (VI) if water is generated as a by-product.

[0053] The substrates whose surface is rendered stainproof according to the present invention can be used as parts of the following articles which are liable to be stained.

[0054] Home electric appliances such as blades of electric fans, doors of electronic ovens, and surface of electric refrigerators; office articles such as contact glass of copying machines, mirrors of OHP main bodies, OHP sheets, key boards, telephones and desks; home articles such as glass, doors of cupboards, mirrors, windowpanes, lampshades and chandeliers; architectural members such as show windows, telephone boxes and glass of water baths; automotive parts such as glass and paint films on bodies of vehicles; personal ornaments such as frames of a pair of spectacles, glass of underwater cameras, goggles, helmets and cover glass of dial plates of watches; play goods such as glass of pinball tables, trumps and mahjong tiles; and paint film surfaces of articles such as furniture and piano.

[0055] Ornament pieces such as necktie pins, necklaces and pierced earring; metal or plated articles such as faucets of water supply, metal wind instruments, wooden wind instruments, golf clubs, door handles, dumbbells and knives; stone articles such as gravestones, go game stone pieces and marbles; paper products such as wall paper, paper for sliding partitions, books, posters and photographs; and leather products such as wallets, shoes, bags, wrist watch bands and baseball gloves.

#### PREFERRED EMBODIMENT OF THE INVENTION

[0056] In order to further illustrate the present invention, the following specific examples are given. It should be understood that the following examples do not limit the present invention.

#### Example 1

[0057] A 500 mL four-neck flask equipped with a stirrer, dropping funnel, refluxing condenser and thermometer was

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charged with a solution prepared by dissolving 180 g (0.5 mol) of  $C_3F_7-(OCF_2CF_2CF_2)_a-OCF_2CF_2COF$  (having a number-average molecular weight of 3,600) in 500 mL of dichloropentafluoropropane (HCFC-225) and the atmosphere of the system was sufficiently replaced with nitrogen. Then, under a nitrogen gas stream, 90 g (0.5 mol) of hydroxypropyltrimethoxysilane (AY-43-024 manufactured by Toray Dow Corning Co., Ltd.) was added dropwise from a dropping funnel to the flask.

[0058] Upon completion of the addition, the temperature of the system was raised to 50°C and the reaction was kept for 4 hours. After the reaction, 250 g of a fluorine-containing polymer  $[C_3F_7-(OCF_2CF_2CF_2)_a-CF_2CF_2COOCH_2CH_2CH_2Si(OCH_3)_3]$  was obtained by removing the solvent HCFC-225 under a reduced pressure.

[0059] According to the infrared absorption spectrum of the reaction product, the absorption ( $1890\text{ cm}^{-1}$ ) due to  $-COF$  of the starting material disappeared completely and new absorption ( $1780\text{ cm}^{-1}$ ) due to  $-COO-$  of ester was found.

[0060] The results of the elemental analysis of the fluorine-containing polymer were as follows.

	C	F	O	H	Si
Calculated (%)	22.7	65.2	10.9	0.4	0.8
Found (%)	21.5	65.8	11.3	0.5	0.9

## Example 2

[0061] A 500 mL four-neck flask equipped with a stirrer, dropping funnel, refluxing condenser and thermometer was charged with a solution prepared by dissolving 100 g (0.5 mol) of  $C_3F_7-(OCF_2CF_2CF_2)_a-OCF_2CF_2CH_2OH$  (having a number-average molecular weight of 2,000) in 500 mL of dichloropentafluoropropane (HCFC-225) and with 1 mL of trifluoroboron ether as a catalyst. The atmosphere of the system was sufficiently replaced with nitrogen. Then, under a nitrogen gas stream, 100 g (0.5 mol) of  $\gamma$ -glycidoxypropyltrimethoxysilane (SH6040 manufactured by Toray Dow Corning Co., Ltd.) was added dropwise from a dropping funnel to the flask.

[0062] Upon completion of the addition, the temperature of the system was raised to 50°C and the reaction was kept for 4 hours. After the reaction, 175 g of a fluorine-containing polymer  $[C_3F_7-(OCF_2CF_2CF_2)_a-OCF_2CF_2CH_2-OCH_2CH(OH)CH_2OCH_2CH_2CH_2Si(OCH_3)_3]$  was obtained by removing the solvent HCFC-225 under a reduced pressure.

[0063] According to the infrared absorption spectrum of the reaction product, the absorption ( $795\text{ cm}^{-1}$ ,  $820\text{ cm}^{-1}$ ) due to epoxy of the starting material disappeared completely.

[0064] The results of the elemental analysis of the fluorine-containing polymer were as follows.

	C	F	O	H	Si
Calculated (%)	22.4	61.1	12.3	1.0	1.2
Found (%)	23.9	61.4	12.5	0.9	1.3

## Example 3

[0065] A 500 mL four-neck flask equipped with a stirrer, dropping funnel, refluxing condenser and thermometer was charged with a solution prepared by dissolving 100 g (0.5 mol) of  $HOCH_2CF_2-(OCF_2CF_2)_b-(OCF_2)_c-OCF_2CH_2OH$  (having a number-average molecular weight of 2,000) in 500 mL of m-xylene hexafluoride and with 1 mL of trifluoroboron ether as a catalyst. The atmosphere of the system was sufficiently replaced with nitrogen. Then, under a nitrogen gas stream, 200 g (1 mol) of  $\gamma$ -glycidoxypropyltrimethoxysilane (SH6040 manufactured by Toray Dow Corning Co., Ltd.) was added dropwise from the dropping funnel to the flask.

[0066] Upon completion of the addition, the temperature of the system was raised to 50°C and the reaction was kept for 4 hours. After the reaction, 250 g of a fluorine-containing polymer  $[(H_3CO)_3SiCH_2CH_2CH_2OCH_2CH(OH)CH_2OCH_2CF_2-(OCF_2CF_2)_b-(OCF_2)_c-OCF_2CH_2-OCH_2CH(OH)CH_2OCH_2CH_2CH_2Si(OCH_3)_3]$  was obtained by removing the solvent m-xylene hexafluoride under a reduced pressure.

[0067] According to the infrared absorption spectrum of the reaction product, the absorption ( $795\text{ cm}^{-1}$ ,  $820\text{ cm}^{-1}$ )

due to epoxy of the starting material disappeared completely.

[0068] The results of the elemental analysis of the fluorine-containing polymer were as follows.

	C	F	O	H	Si
Calculated (%)	26.1	51.9	17.5	1.9	2.6
Found (%)	25.5	51.5	18.4	1.9	2.7

#### Examples 4 to 5 and Comparative Example 1

[0069] The polymer obtained in Example 1 (Example 4), the polymer obtained in Example 2 (Example 5), and a commercially available, fluorine-containing silane coupling agent,  $C_8F_{17}CH_2CH_2Si(OCH_3)_3$ , KBM7803 manufactured by Shin-Etsu Chemical Co., Ltd. (hereinafter simply referred to as "commercial product") (Comparative Example 1) were dissolved, respectively, in perfluorohexane to prepare 0.1% by weight solutions. These solutions were used as stainproof treatment liquids.

[0070] Meanwhile, glass plates for use as substrates were washed with water and thereafter well washed with methanol and acetone. The glass plates thus prepared were immersed in the treatment liquid for 10 seconds, taken out of the liquid, and then dried for 60 minutes at room temperature. Then, the glass plates were subjected to ultrasonic cleaning in perfluorohexane for 5 minutes to remove excessive stainproof treatment components. Then, the glass plates were dried in air and subjected to various measurements.

#### Example 6

[0071] The polymer obtained in Example 3 was dissolved in perfluorohexane to prepare a 0.1% by weight solution. This solution was used as a stainproof treatment liquid.

[0072] Meanwhile, a glass plate for use as a substrate was washed with water and thereafter well washed with methanol and acetone. The glass plate thus prepared was immersed in the treatment liquid for 10 seconds, taken out of the treatment liquid, and then dried for 60 minutes at room temperature. Then, the glass plate was subjected to ultrasonic cleaning in perfluorohexane for 5 minutes to remove excessive stainproof treatment components. Then, the glass plate was dried in air and subjected to various measurements.

#### Example 7

[0073]  $\gamma$ -(2-aminoethyl)aminopropyltrimethoxysilane (SH6020 manufactured by Toray Dow Corning Co., Ltd.) (silane coupling agent) was dissolved in acetone to prepare a 0.1% by weight solution.

[0074] Meanwhile, a glass plate for use as a substrate was washed with water and thereafter well washed with methanol and acetone. The glass plate thus prepared was immersed in the solution for 1 minute, taken out of the solution, and then dried for 1 hour at room temperature.

[0075] Next,  $C_3F_7-(OCF_2CF_2CF_2)_n-OCF_2CF_2COOH$  (having a number-average molecular weight of 5,000) was dissolved in perfluorohexane to prepare a 0.5% by weight solution. In this solution, the glass plate, which had been treated with the silane coupling agent, was immersed for 1 minute. The glass plate was then taken out of the solution and dried for 1 hour at 100°C.

[0076] Then, the glass plate was subjected to ultrasonic cleaning in perfluorohexane for 5 minutes to remove excessive stainproof treatment components. Then, the glass plate was dried in air and subjected to various measurements.

#### Example 8

[0077]  $\gamma$ -mercaptopropyltrimethoxysilane (SH6062 manufactured by Toray Dow Corning Co., Ltd.) (silane coupling agent) was dissolved in acetone to prepare a 0.1% by weight solution.

[0078] Meanwhile, a glass plate for use as a substrate was washed with water and thereafter well washed with methanol and acetone. The glass plate thus prepared was immersed in the solution for 1 minute, taken out of the solution, and then dried for 1 hour at room temperature.

[0079] Next,  $C_3F_7-(OCF_2CF_2CF_2)_n-OCF_2CF_2COOH$  (having a number-average molecular weight of 2,000) was dissolved in perfluorohexane to prepare a 0.5% by weight solution. In this solution, the glass plate, which had been treated with the silane coupling agent, was immersed for 1 minute. The glass plate was then taken out of the solution

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and dried for 1 hour at 100°C.

[0080] Then, the glass plate was subjected to ultrasonic cleaning in perfluorohexane for 5 minutes to remove excessive stainproof treatment components. Then, the glass plate was dried in air and subjected to various measurements.

[0081] The properties of the glass plates treated were evaluated according to the following methods.

(1) As to the adhesion of fingerprints, the glass plate samples were stained with fingerprints and the ease of the fingerprints adherence were visually inspected.

○ : the amount of adhered fingerprint is slight so that the adhered fingerprint is not apparent.

X: the amount of fingerprint is the same as that on an untreated glass plate.

Δ: difficult to give a clear rating between the two described above

(2) As to the ease in fingerprint removal, the surfaces of the samples used in the evaluation of the adhesion of fingerprints were wiped by one cycle of reciprocating movement of KIM Wipe (manufactured by Jujo Kimberley Co., Ltd.) and the ease in removal of the fingerprints was visually inspected.

○ : fingerprints can be completely removed.

Δ: fingerprint wiping marks remains.

X: fingerprints wiping marks extend and removal of the marks is difficult.

(3) Contact angle with water was measured using a contact angle meter (model CA-DT manufactured by Kyowa Kaimem Kagaku Kikai Co., Ltd.).

[0082] The results are given below.

	Adhesion of fingerprints	Ease in fingerprint removal	Contact angel with water
Example 4	○	○	114°
Example 5	○	○	110°
Example 6	○	○	109°
Example 7	○	○	115°
Example 8	○	○	109°
Comparative Example 1	Δ	X	109°

Example 9 and Comparative Example 2

[0083] The polymer obtained in Example 1 (Example 9) and a commercially available product (Comparative Example 2) were dissolved, respectively, in perfluorohexane to prepare 0.1% by weight solutions.

[0084] Meanwhile, aluminum plates for use as substrates (each having a thickness of 0.5 mm and in compliance with JIS H 4000 (A1050P)) were washed with water and thereafter well washed with methanol and acetone.

[0085] The aluminum plates thus prepared were immersed in the solutions for 10 seconds, taken out of the solution, and then dried for 60 minutes at room temperature. Then, the aluminum plates were subjected to ultrasonic cleaning in perfluorohexane for 5 minutes to remove excessive stainproof treatment components. Then, the aluminum plates were dried in air and subjected to various measurements.

Comparative Example 3

[0086] The aluminum plate in Example 9 was used without being treated.

Example 10

[0087]  $\gamma$ -(2-aminoethyl)aminopropyltrimethoxysilane (SH6020 manufactured by Toray Dow Corning Co., Ltd.) (silane

coupling agent) was dissolved in acetone to prepare a 0.1% by weight solution.

[0088] Meanwhile, an aluminum plate for use as a substrate (having a thickness of 0.5 mm and in compliance with JIS H 4000 (A1050P)) was washed with water and thereafter well washed with methanol and acetone.

[0089] The aluminum plate thus prepared was immersed in the solution for 1 minute, taken out of the solution, and then dried for 60 minutes at room temperature.

[0090] Next,  $C_3F_7-(OCF_2CF_2CF_2)_a-OCF_2CF_2COOH$  (having a number-average molecular weight of 5,000) was dissolved in perfluorohexane to prepare a 0.5% by weight solution. In this solution, the aluminum plate, which had been treated with the silane coupling agent, was immersed for 1 minute. The aluminum plate was then taken out of the solution and dried for 1 hour at 100°C.

[0091] Then, the aluminum plate was subjected to ultrasonic cleaning in perfluorohexane for 5 minutes to remove excessive stainproof treatment components. Then, the aluminum plate was dried in air and subjected to the following durability test.

[0092] The durability test was conducted in the following manner. The same sample surface was wiped by 100 cycles of reciprocating movement of KIM Wipe (manufactured by Jujo Kimberley Co., Ltd.) under a load of 900 g by using an electric clock meter (416-TM1 model manufactured by Yasuda Seiki Seisakusho Co., Ltd.). The contact angles with water before and after the wiping operations were measured. The contact angle with water was measured using a contact angle meter (model CA-DT manufactured by Kyowa Kaimem Kagaku Kikai Co., Ltd.).

[0093] The results are given below.

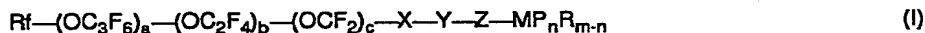
	Contact angle with water before wiping operation	Contact angle with water after wiping operation
Example 9	112°	107°
Example 10	115°	111°
Comparative Example 2	111°	98°
Comparative Example 3	88°	—

## EFFECTS OF THE INVENTION

[0094] When used as a stainproof coating layer, the fluorine-containing polymer of the present invention provides a stainproof substrate having an excellent stainproof property against particularly fingerprints. Accordingly, the fluorine-containing polymer of the present invention can be suited for use in a variety of substrates such as glass, resins, metals, ceramics, wood, porcelain, stone and leather.

## Claims

1. A fluorine-containing polymer having a number-average molecular weight of 500 to 100,000 and represented by the following general formula (I):



wherein Rf is a perfluoroalkyl group;

a, b and c independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of a, b and c is at least 1;

X is a group represented by the formula:  $-(O)_d-(CF_2)_e-(CH_2)_f$  (where d, e and f independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of e and f is at least 1, and the presence order of the bracketed repeating units is arbitrary in the formula);

Y is a divalent polar group;

Z is a group represented by the formula:  $-(CH_2)_g$  (where g is 0 or a number equal to or greater than 1);

M is a metal atom;

R is a hydrocarbon group;

P is a hydrolyzable polar group;

m is an integer equal to "(valency of M) - 1";

n is an integer of 1 to m;

with the proviso that the presence order of the bracketed repeating units is arbitrary in the formula;  $-\text{OC}_3\text{F}_6-$  represents either  $-\text{OCF}_2\text{CF}_2\text{CF}_2-$  or  $-\text{OCF}(\text{CF}_3)\text{CF}_2-$ ; and  $-\text{OC}_2\text{F}_4-$  represents either  $-\text{OCF}_2\text{CF}_2-$  or  $-\text{OCF}(\text{CF}_3)-$ .

2. A fluorine-containing polymer having a number-average molecular weight of 500 to 100,000 and represented by the following general formula (II):



wherein a, b and c independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of a, b and c is at least 1;

X is a group represented by the formula:  $-(\text{O})_d-(\text{CF}_2)_e-(\text{CH}_2)_f-$  (where d, e and f independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of e and f is at least 1, and the presence order of the bracketed repeating units is arbitrary in the formula);

Y is a divalent polar group;

Z is a group represented by the formula:  $-(\text{CH}_2)_g-$  (where g is 0 or a number equal to or greater than 1);

M is a metal atom;

R is a hydrocarbon group;

P is a hydrolyzable polar group;

m is an integer equal to "(valency of M) - 1";

n is an integer of 1 to m;

with the proviso that the presence order of the bracketed repeating units is arbitrary in the formula;  $-\text{OC}_3\text{F}_6-$  represents either  $-\text{OCF}_2\text{CF}_2\text{CF}_2-$  or  $-\text{OCF}(\text{CF}_3)\text{CF}_2-$ ; and  $-\text{OC}_2\text{F}_4-$  represents either  $-\text{OCF}_2\text{CF}_2-$  or  $-\text{OCF}(\text{CF}_3)-$ .

3. A fluorine-containing polymer having a number-average molecular weight of 500 to 100,000 and represented by the following general formula (III):



wherein Rf is a perfluoroalkyl group;

a is a number equal to or greater than 1;

X is a group represented by the formula:  $-(\text{O})_d-(\text{CF}_2)_e-(\text{CH}_2)_f-$  (where d, e and f independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of e and f is at least 1, and the presence order of the bracketed repeating units is arbitrary in the formula);

Y is a divalent polar group;

Z is a group represented by the formula:  $-(\text{CH}_2)_g-$  (where g is 0 or a number equal to or greater than 1);

R is a hydrocarbon group;

P is a hydrolyzable polar group; and

n is an integer of 1 to 3.

4. A stainproof substrate comprising a substrate, and a layer which is provided on said substrate and composed of the fluorine-containing polymer of Claim 1.
5. A stainproof substrate comprising a substrate and a layer which is provided on said substrate and composed of the fluorine-containing polymer of Claim 2.
6. A stainproof substrate comprising a substrate and a layer which is provided on said substrate and composed of the fluorine-containing polymer of Claim 3.
7. A process for preparing the stainproof substrate of Claim 4 comprising the steps of forming on a surface of the substrate a layer of a compound represented by the following general formula (IV):



wherein Z is a group represented by the formula :  $-(CH_2)_g-$  (where g is 0 or a number equal to or greater than 1);

M is a metal atom;

R is a hydrocarbon group;

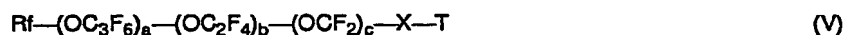
P is a hydrolyzable polar group;

m is an integer equal to "(valency of M) - 1";

n is an integer of 1 to m; and

Q is a polar group;

and thereafter reacting said layer with a compound represented by the following general formula (V):



wherein Rf is a perfluoroalkyl group;

a, b and c independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of a, b and c is at least 1;

X is a group represented by the formula :  $-(O)_d-(CF_2)_e-(CH_2)_f-$  (where d, e and f independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of e and f is at least 1, and the presence order of the bracketed repeating units is arbitrary in the formula); and

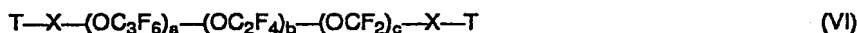
T is a polar group;

with the proviso that the presence order of the bracketed repeating units is arbitrary in the formula;  $-OC_3F_6-$  represents either  $-OCF_2CF_2CF_2-$  or  $-OCF(CF_3)CF_2-$ ; and  $-OC_2F_4-$  represents either  $-OCF_2CF_2-$  or  $-OCF(CF_3)-$ .

8. A process for preparing the stainproof substrate of Claim 5 comprising the steps of forming on the surface of the substrate a layer of a compound represented by the following general formula (IV):



wherein Q, Z, M, P, R, m and n are the same as defined in the formula (IV) of Claim 7, and thereafter reacting said layer with a compound represented by the following general formula (VI):



wherein Rf is a perfluoroalkyl group;

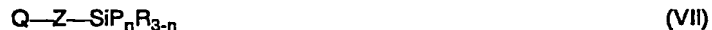
a, b and c independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of a, b and c is at least 1;

X is a group represented by the formula :  $-(O)_d-(CF_2)_e-(CH_2)_f-$  (where d, e and f independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of e and f is at least 1, and the presence order of the bracketed repeating units is arbitrary in the formula); and

T is a polar group;

with the proviso that the presence order of the bracketed repeating units is arbitrary in the formula;  $-OC_3F_6-$  represents either  $-OCF_2CF_2CF_2-$  or  $-OCF(CF_3)CF_2-$ ; and  $-OC_2F_4-$  represents either  $-OCF_2CF_2-$  or  $-OCF(CF_3)-$ .

9. A process for preparing the stainproof substrate of Claim 6 comprising the steps of forming on the surface of the substrate a layer of a compound represented by the following general formula (VII):



wherein Z is a group represented by the formula :  $-(CH_2)_g-$  (where g is 0 or a number equal to or greater than 1);

R is a hydrocarbon group;

P is a hydrolyzable polar group;

n is an integer of 1 to 3; and

Q is a polar group, and thereafter reacting said layer with a compound represented by the following general



formula (VIII):



wherein Rf is a perfluoroalkyl group;

a is a number equal to or greater than 1;

X is a group represented by the formula:  $-(\text{O})_d-(\text{CF}_2)_e-(\text{CH}_2)_f-$  (where d, e and f independently represent 0 or a number equal to or greater than 1, with the proviso that the sum of e and f is at least 1, and the presence order of the bracketed repeating units is arbitrary in the formula); and

T is a polar group.

10. A process for preparing the fluorine-containing polymer of Claim 1 comprising the steps of carrying out a reaction between a compound represented by the general formula (IV):



wherein Q, Z, M, P, R, m and n are the same as defined in the formula (IV) of Claim 7, and a compound represented by the following general formula (V):

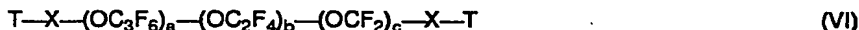


wherein Rf, T, X, a, b and c are the same as defined in the formula (V) of Claim 7.

11. A process for preparing the fluorine-containing polymer of Claim 2 comprising the steps of carrying out a reaction between a compound represented by the general formula (IV):



wherein Q, Z, M, P, R, m and n are the same as defined in the formula (IV) of Claim 7, and a compound represented by the following general formula (VI):



wherein T, X, a, b and c are the same as defined in the formula (VI) of Claim 8.

12. A process for preparing the fluorine-containing polymer of Claim 3 comprising the steps of carrying out a reaction between a compound represented by the general formula (VII):



wherein Q, Z, P, R and n are the same as defined in the formula (VII) of Claim 9, and a compound represented by the following general formula (VIII):



wherein T, Rf, X and a are the same as defined in the formula (VIII) of Claim 9.

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP98/01948

<b>A. CLASSIFICATION OF SUBJECT MATTER</b> Int.Cl. <sup>6</sup> C08G65/32, C09D171/02 // C08G77/46, C09D183/12 According to International Patent Classification (IPC) or to both national classification and IPC		
<b>B. FIELDS SEARCHED</b> Minimum documentation searched (classification system followed by classification symbols) Int.Cl. <sup>6</sup> C08G65/00-65/48, C09D171/00-171/14 Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) WPI/L (QUESTEL)		
<b>C. DOCUMENTS CONSIDERED TO BE RELEVANT</b>		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	JP, 5-117283, A (Mitsubishi Rayon Co., Ltd.), May 14, 1993 (14. 05. 93), Claims ; Par. No. [0013] (Family: none)	1, 3, 4, 6, 7, 9 2, 5, 8
X	JP, 5-279499, A (Asahi Glass Co., Ltd.), October 26, 1993 (26. 10. 93), Claims ; Par. Nos. [0018] to [0020], [0026] to [0030], [0039], [0050] & EP, 513690, A2 & US, 5314731, A	1, 3 2, 4-9
PX	JP, 9-137027, A (Shin-Etsu Chemical Co., Ltd.), May 27, 1997 (27. 05. 97), Claims ; Par. Nos. [0006], [0043], [0044] & US, 5705591, A	2 5, 8 1, 3, 4, 6, 7, 9
PA		
X	JP, 62-48724, A (Minnesota Mining and Manufacturing Co.), March 3, 1987 (03. 03. 87), Claims & US, 3810874, A	2 1, 3-9
A		
<input checked="" type="checkbox"/> Further documents are listed in the continuation of Box C. <input type="checkbox"/> See patent family annex.		
* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed		"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "Z" document member of the same patent family
Date of the actual completion of the international search July 21, 1998 (21. 07. 98)		Date of mailing of the international search report August 4, 1998 (04. 08. 98)
Name and mailing address of the ISA/ Japanese Patent Office		Authorized officer
Facsimile No.		Telephone No.

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EP 0 978 524 A1

INTERNATIONAL SEARCH REPORT

International application No.  
PCT/JP98/01948

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	JP, 6-256500, A (Shin-Etsu Chemical Co., Ltd.), September 13, 1994 (13. 09. 94), Claims (Family: none)	1-9
A	JP, 7-505399, A (E.I. Du Pont de Nemours & Co.), June 15, 1995 (15. 06. 95), Claims & WO, 9320084, A1 & EP, 629205, A1	1-9

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